# FINAL DRAFT PRELIMINARY ASSESSMENT BRINKMANN INSTRUMENTS, INC. WESTBURY, NEW YORK

PREPARED UNDER

TECHNICAL DIRECTIVE DOCUMENT NO. 02-8901-19
CONTRACT NO. 68-01-7346

**FOR THE** 

ENVIRONMENTAL SERVICES DIVISION
U.S. ENVIRONMENTAL PROTECTION AGENCY

**APRIL 14, 1989** 

NUS CORPORATION SUPERFUND DIVISION

**SUBMITTED BY:** 

RICHARD L. FEINBERG

PROJECT MANAGER

MICHAELBAUMAN

SITE MANAGER

**REVIEWED/APPROVED BY:** 

RONALD M. NAMAN FIT OFFICE MANAGER

299947

# POTENTIAL HAZARDOUS WASTE SITE PRELIMINARY ASSESSMENT

### **PART I: SITE INFORMATION**

1.	Site Name/Alias _	<u>Brinkmann Insti</u>	ruments inc.		•	•
	Street <u>Cantia</u>	gue Rock Road			•	
	City Westburn			State	NY	Zip <u>11590</u>
2.	County Nassau	···		County Code	059	Cong. Dist. 04
3	EPA ID No. NY	D002054351				
4.	Latitude <u>40° 46′</u>	35" N		Longitude	73° 33′ 12″ W	1
	USGS Quad					
5.	Owner Brinkmann					
	Street Cantiague I					
	City Westbury		<del></del>	State	VY	Zip <u>11590</u>
6.	Operator Brinkma	ann instruments	Inc.	Tel. No. <u>(516</u>	334-7500	
	Street Cantiaque	Rock Road	·			
	City Westbury	·		State NY		Zip <u>11590</u>
7.	Type of Ownershi	p				. •
	<b>⊠</b> Private	☐ Federal	☐ State	ı		
	□ County	☐ Municipal	☐ Unkn	own	☐ Other	
8.	Owner/Operator N	lotification on F	ile			
	☑ RCRA 3001	Date <u>8/1</u>	5/80	CERCLA 103c	Date	
	☐ None	☐ Unkno	own			
9.	Permit Information	1				
	Permit None	Permit No.	Date Issued	Expirat	ion Date	Comments
10.	Site Status		•			-
	<b>⊠</b> Active	☐ Inactive		Unknown		
1.	Years of Operation	1974	to	Present		
<b>2.</b>	Identify the types above- or below-g waste unit number	round tanks or	containers, lan	d treatment. 4	etc.) on site	les, stained soil, Initiate as many
	Waste Unit I 1			aste Unit Type		·
3.	Information availa	ble from				
	Contact Amy Bro	chu	Agency U.S.	EPA	Tel. No	. (201) 906-6802
	Preparer Michael B	auman	Agency NUS	Corporation	Date_	

#### PART II: WASTE SOURCE INFORMATION

		•
For each of the waste unit	identified in Part I. co	mplete the following seven items.

Waste Unit No. 1 - Septic Tank

1. Identify the RCRA permit status, if applicable, and the age of the waste unit.

Neither the septic tank nor any other part of the facility has a RCRA permit. In August of 1980, Brinkmann filed a Notification of Hazardous Waste Activity form at the U.S. Environmental Protection Agency (EPA). In November of 1980, Brinkmann filed an application for a RCRA permit. In August of 1981, Brinkmann withdrew its application for a RCRA permit because it no longer handled hazardous substances on site. The Brinkmann facility became active in 1974; however, information to establish whether hazardous substances have been handled continuously from 1974 to 1981 was not available.

2. Describe the location of the waste unit and identify clearly on the site map.

The waste unit is an underground septic tank located outside the Brinkmann building, within the property boundary. A drain line connects the septic tank to a sink in the laboratory. The exact location of the septic tank is not known; however, a map in Ref. No. 3, which has no north arrow, depicts the septic tank.

3. Identify the size or quantity of the waste unit (e.g., area or volume of a landfill or surface impoundment, number and capacity of drums or tanks). Specify the quantity of hazardous substances in the waste unit.

The design capacity of the septic tank is 1000 gallons. The exact quantity of hazardous substances in the waste unit is unknown. However, the RCRA Hazardous Waste Application indicates that approximately 70 gallons of solvents were handled on site per year. It is not known how much of these solvents reached the septic tank.

4. Identify the physical state(s) of the waste type(s) as disposed of in the waste unit. The physical state(s) should be categorized as follows: solid, powder or fines, sludge, slurry, liquid, or gas.

The physical state of the solvents handled on site is liquid. When these solvents were discharged to the septic tank, they were diluted with water.

5. Identify specific hazardous substance(s) known or suspected to be present in the waste unit.

Acetone, chloroform, 1,2-dichloroethane, ethyl acetate, methanol, and dichloromethane were known to be handled on site and potentially disposed of into the septic tank.

6. Describe the containment of the waste unit as it relates to contaminant migration via groundwater, surface water, and air.

The hazardous substances reaching the septic tank are assumed to have been released to groundwater. A release to surface water or air is unlikely because the septic tank is underground.

7. Identify any miscellaneous spills, dumping, etc. on site; describe the materials and identify their locations on site.

No miscellaneous spills or incidents of dumping were noted in available background information.

Ref. Nos. <u>1, 2, 3, 4, 5, 18</u>

#### PART III: HAZARD ASSESSMENT

#### **GROUNDWATER ROUTE**

1. Describe the likelihood of a release of contaminant(s) to the groundwater as follows: observed, alleged, potential, or none. Identify the contaminant(s) detected or suspected, and provide a rationale for attributing the contaminant(s) to the facility.

Until 1981, solvents from this facility were disposed of in laboratory sinks and flushed into drains leading to a septic tank. It is likely that a release to groundwater occurred when the septic tank was used in this manner. The potential contaminants are acetone, chloroform, 1,2-dichloroethane, ethyl acetate, methanol, and dichloromethane as listed in the U.S. EPA Notification of Hazardous Waste Activity. These compounds were all used in the process of manufacturing the medical test kits this company produced.

Ref. Nos. 2, 3, 4, 5

2. Describe the aquifer of concern; include information such as depth, thickness, geologic composition, permeability, overlying strata, confining layers, interconnections, discontinuities, depth to water table, groundwater flow direction.

Directly beneath the site, the Magothy aquifer is the aquifer of concern, and its surface lies at an elevation of approximately 0 feet mean sea level (MSL). This clay, silt, sandy clay, and fine to medium sand formation is approximately 435 feet thick. A large quantity of clay in the upper portion of the Magothy causes water to become confined and may even cause the formation to display artesian conditions with depth. The Magothy displays moderate to high permeability. Wells screened in the lower part of the aquifer can produce as much as 1,400 gal/min. The Magothy displays a high degree of hydraulic continuity with the overlying upper glacial aquifer but can vary from location to location throughout Long Island. The upper glacial aquifer, which lies from the surface down to a depth of approximately 158 feet, consists of glacial till which is unsorted clay, sand, gravel, and boulders. It may also contain outwash deposits of statified brown sand and gravel, and lacustrine or marine deposits consisting of clay, silt, and sand which can be fossiliferous. The glacial till may cause localized, perched water conditions and retard downward percolation of precipitation. However, the outwash deposits are highly permeable. In general, this formation contains the water table for Long Island and recharges all underlying aquifers. Beneath the site, the water table is at a depth of approximately 68 feet or 90 feet above MSL. Groundwater is moving in a southward direction.

Ref. No. 6

3. Is a designated sole source aquifer within 3 miles of the site?

Three sole source aquifers, the upper glacial aquifer, the Magothy aquifer, and the Lloyd aquifer, underlie the site.

Ref. No. 7

4. What is the depth from the lowest point of waste disposal/storage to the highest seasonal level of the saturated zone of the aquifer of concern?

The lowest point of waste disposal is assumed to be 6 feet, because the septic tank is underground but information to determine its lowest point is not available. The water table in the upper glacial aquifer, which is hydraulically connected to the Magothy aquifer, is at approximately 90 feet above MSL, and the facility elevation is approximately 160 feet MSL. Therefore, the depth from the lowest point of waste disposal to the aquifer of concern is approximately 60 feet.

Ref. Nos. 3, 6, 8

5. What is the permeability value of the least permeable intervening strata between the ground surface and the aguifer of concern?

The least permeable zone between the surface and the upper glacial aquifer is estimated to be greater than 10<sup>-3</sup> cm/sec.

Ref. Nos. 6, 9

6. What is the net precipitation for the area?

The net precipitation in the vicinity of the site is approximately 15 inches.

Ref. No. 9

7. Identify uses of groundwater within 3 miles of the site (i.e., private drinking source, municipal source, commercial, industrial, irrigation, unusable).

Five water districts have municipal supply wells located within 3 miles of the site.

Ref. Nos. 10, 11, 12, 13, 14

8. What is the distance to and depth of the nearest well that is currently used for drinking or irrigation purposes?

Distance Approximately 600 feet east

Depth 535 feet

Ref. No. 13

9. Identify the population served by the aquifer of concern within a 3-mile radius of the site.

Five water districts have 43 municipal supply wells located within 3 miles of the site that draw from the Magothy aquifer. These water districts serve approximately 160,000 people.

Ref. Nos. 10, 11, 12, 13, 14, 15

#### **SURFACE WATER ROUTE**

10. Describe the likelihood of a release of contaminant(s) to surface water as follows: observed, alleged, potential, or none. Identify the contaminant(s) detected or suspected, and provide a rationale for attributing the contaminants to the facility.

There is no potential for a release to surface water because the contaminants are underground. Further, a surface water migration pathway does not exist. Storm water runoff drains from the site via storm drains, and enters a recharge basin where it infiltrates the ground. Because there is no surface water migration route, the surface water route will not be considered.

Ref. Nos. 1, 8, 16

11. What is the facility slope in percent? (Facility slope is measured from the highest point of deposited hazardous waste to the most downhill point of the waste area or to where contamination is detected.)

**Not Applicable** 

12. What is the slope of the intervening terrain in percent? (Intervening terrain slope is measured from the most downhill point of the waste area to the probable point of entry to surface water).

Not Applicable

13. What is the 1-year 24-hour rainfall?

Not Applicable

14. What is the distance to the nearest downslope surface water? Measure the distance along a course that runoff can be expected to follow.

Not Applicable

15. Identify uses of surface waters within 3 miles downstream of the site (i.e., drinking, irrigation, recreation, commercial, industrial, not used).

**Not Applicable** 

16. Describe any wetlands, greater than 5 acres in area, within 2 miles downstream of the site. Include whether it is a freshwater or coastal wetland.

**Not Applicable** 

17. Describe any critical habitats of federally-listed endangered species within 2 miles of the site along the migration path.

**Not Applicable** 

18. What is the distance to the nearest sensitive environment along or contiguous to the migration path (if any exist within 2 miles)?

**Not Applicable** 

19. Identify the population served or acres of food crops irrigated by surface water intakes within 3 miles downstream of the site and the distance to the intake(s).

Not Applicable

20. What is the state water quality classification of the water body of concern?

Not Applicable

21. Describe any apparent biota contamination that is attributable to the site.

**Not Applicable** 

#### **AIR ROUTE**

22. Describe the likelihood of a release of contaminant(s) to the air as follows: observed, alleged, potential, none. Identify the contaminant(s) detected or suspected, and provide a rationale for attributing the contaminant(s) to the facility.

There is no potential for a release to the air because the solvents were disposed of in an underground septic tank and the use of these solvents was discontinued in 1981.

Ref. Nos. 3, 4, 5

23. What is the population within a 4-mile radius of the site?

The population within 4 miles of the site is approximately 180,800.

Ref. No.17

#### FIRE AND EXPLOSION

24. Describe the potential for a fire or explosion to occur with respect to the hazardous substance(s) known or suspected to be present on site. Identify the hazardous substance(s) and the method of storage or containment associated with each.

No fire or explosive condition due to the hazardous substances is likely to exist at the site because the quantities of solvents used on site were small, and because the solvents were diluted with water and drained into a septic tank. These substances have not been used on site since 1981.

Ref. Nos. 1, 3, 17

25. What is the population within a 2-mile radius of the hazardous substance(s) at the facility?

The population within 2 miles of the site is approximately 43,200.

**Ref. No.17** 

#### **DIRECT CONTACT/ON-SITE EXPOSURE**

26. Describe the potential for direct contact with hazardous substance(s) stored in any of the waste units on site or deposited in on-site soils. Identify the hazardous substance(s) and the accessibility of the waste unit.

Direct contact is unlikely to occur because the hazardous substances were disposed of in an underground septic tank and hazardous substances have not been used on site since 1981.

Ref. Nos. 3, 4, 5

27. How many residents live on a property whose boundaries encompass any part of an area contaminated by the site?

There is no area with documented evidence of contamination other than the septic tank. The septic tank is totally contained on the Brinkmann property.

Ref. No. 3

28. What is the population within a 1-mile radius of the site?

The population within 1 mile of the site is approximately 12,600.

Ref. No. 3

#### PART IV: SITE SUMMARY AND RECOMMENDATIONS

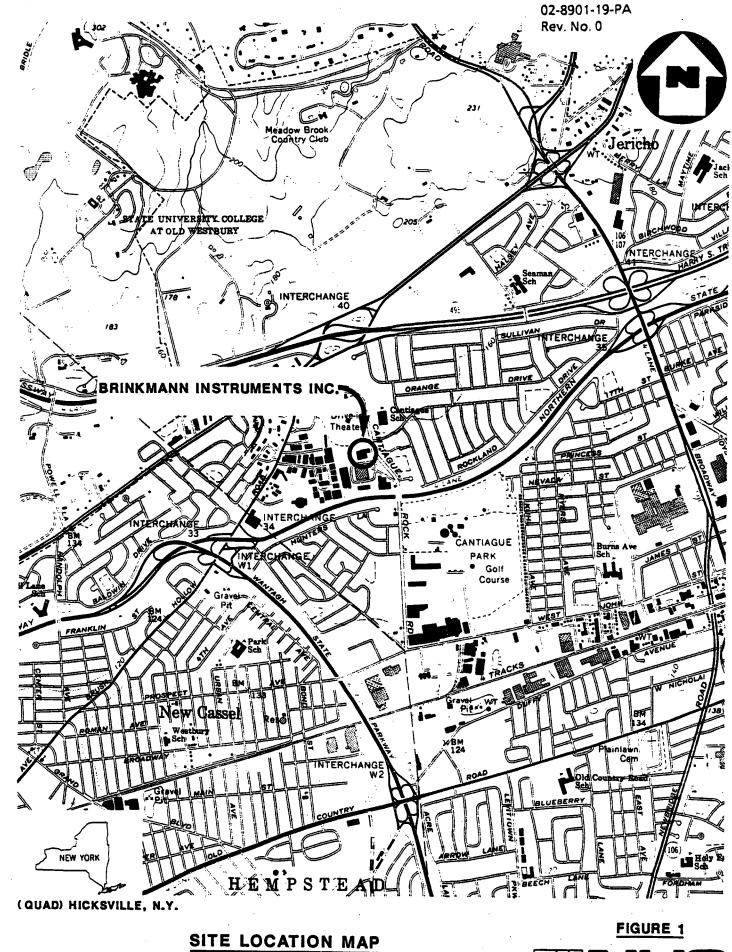
The Brinkmann Instruments Site is located on Cantiague Rock Road in Westbury, NY. Westbury is located on Long Island in a heavily populated section of Nassau County. Brinkmann, a subsidiary of Sybron Corporation, has operated the 10-acre facility on Cantiague Rock Road since 1974 where it imports and distributes scientific instruments.

Prior to 1982 Brinkmann operated a laboratory in the single building on site where waste solvents were produced. An annual amount of approximately 70 gallons of solvents was used in the production of clinical test kits. A preliminary assessment was conducted on this site to assess the potential impact of these solvents. Groundwater is the only potentially affected medium. Solvents in the laboratory migrate to groundwater from the laboratory sink into the septic tank. Storm water runoff in this area is directed to recharge basins and therefore never enters any surface water bodies. There is a recharge basin across the street from a neighboring property. There is no longer a potential for a release to air because use of the solvents was ceased in 1981.

A recommendation is made for NO FURTHER REMEDIAL ACTION PLANNED (NFRAP) at this site. There is no potential for a release to air or to surface water. There is a potential that a limited release to groundwater may have occurred prior to 1982. However, further testing of groundwater is not recommended at this time for the following reasons:

- The total amount of solvents known to have been on site is minimal, and only a portion of this minimal quantity is expected to have been released.
- The last known time a potential release may have occurred was 8 years ago, and the information for identifying groundwater flow dynamics with sufficient accuracy to determine the present-day location of these potential releases is not available.
- There are no known wells in the area that are at a depth where contaminants may potentially be found.

ATTACHMENT 1



BRINKMANN INSTRUMENTS INC., WESTBURY, N.Y.

SCALE: 1'= 2000'



SITE MAP

BRINKMANN INSTRUMENTS INC., WESTBURY, N.Y.

(NOT TO SCALE)





#### EXHIBIT A

PHOTOGRAPH LOG

BRINKMANN INSTRUMENTS INC. WESTBURY, NEW YORK

#### BRINKMANN INSTRUMENTS INC. WESTBURY, NEW YORK JANUARY 11, 1989

#### PHOTOGRAPH INDEX

# ALL PHOTOGRAPHS TAKEN BY PETER VON SCHONDORF

Photo Number	<u>Description</u>	Time		
1P-17	Corner of site from the west showing the fence.	1230		
1P-18	Near the corner of the building from the west.	1232		
1P-19	Side of building from the south.	1235		
1P-20	Building sign from east of site.	1236		



BRINKMANN INSTRUMENTS INC., WESTBURY, NEW YORK



1P-17 January 11, 1989 1230 Corner of site from the west showing the fence.



January 11, 1989 1232 Near the corner of the building from the west.

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BRINKMANN INSTRUMENTS INC., WESTBURY, NEW YORK



1P-19

January 11, 1989 Side of building from the south.

1235



1P-20

January 11, 1989 Building sign from east of site.

ATTACHMENT 2

#### REFERENCES

- 1. Preliminary Assessment Off-Site Reconnaissance Information Reporting Form, Brinkmann Instruments, Inc., TDD No. 02-8901-19, NUS Corp. Region 2 FIT, Edison, NJ, January 10, 1989.
- 2. Notification of Hazardous Waste Activity, U.S. EPA Form No. 0246-EPA-OT, completed by Richard D. Gordon, Brinkmann Instruments, August 15, 1980.
- 3. General Information (Form 1) and Hazardous Waste Permit Application (Form 3), U.S. EPA Form No. 158-R0175, completed by Reiner H. Kopp, November 19, 1980.
- 4. Letter from J.D. Brown, Brinkmann Instruments, to Permit Contact, Permits Adm. Branch, U.S. EPA, August 17, 1981.
- 5. Letter from J.D. Brown, Brinkmann Instruments, to Clinical Customer, June 19, 1981.
- 6. Kilburn, Chabot and Richard K. Krulikas, Hydrogeology and Groundwater Quality of the Northern Part of the Town of Oyster Bay, Nassau County, New York. In 1980, U.S. Geological Survey, Water Resources Investigations Report 85-4051, 1987.
- 7. Federal Register, Vol. 43, No. 120-Wednesday, June 21, 1978, pages 2611 and 26612.
- 8. Three-Mile Vicinity Map, based on U.S. Department of the Interior, Geological Survey Topographic Maps, 7.5 minute series, "Hicksville, N.Y.", 1967, photorevised 1979, and "Freeport, N.Y.", 1969, photorevised 1979.
- 9. Uncontrolled hazardous waste site ranking system, A user's manual, 40 CFR, Part 300, Appendix A, 1986.
- 10. Letter from John J. McCrosson, Hicksville Water District, to E.L. Leonard, NUS Corp., March 22, 1988.
- 11. Letter from Arthur J. Lindon, Village of Westbury Public Works, to Edward L. Leonard, NUS Corp., April 4, 1988.
- 12. Letter from Italo J. Vacchio, Westbury Water District, to E. Leonard, NUS Corp., March 21, 1988.
- 13. Letter from William Evers, Jericho Water District, to Edward Leonard, NUS Corp., May 9, 1988.
- 14. Letter from Howard V. Morgan, Town of Hempstead Department of Water, to Edward L. Leonard, NUS Corp., April 6, 1988.
- 15. New York State Atlas of Community Water System Sources, New York State Department of Health, Division of Environmental Protection, Bureau of Public Water Supply Protection, 1982.
- 16. Seaburn, G.E. and D.A. Aronson, Catalog of Recharge Basins on Long Island, New York, in 1969, U.S. Geological Survey Bulletin 70, 1973.
- 17. General Sciences Corporation, Graphical Exposure Modeling Systems (GEMS). Landover, Maryland, 1986.
- 18. Code of Federal Regulations, revised as of July 1, 1988, Volume 40, Part 261.33.

REFERENCE NO. 1

#### PRELIMINARY ASSESSMENT OFF SITE RECONNAISSANCE INFORMATION REPORTING FORM

Date: 11089	
Site Name: Brinkmann Intaments Inc	. TDD: 02-8901-19
Site Address: <u>Cantingue Roch Road</u> Street, Box, etc.	
Westbury Town	
Nassau County	
N Y State	
NUS Personnel: Name	Discipline
Pete von Schondorf	Ceologist
G. Hannay	Biologis+
3. Dietz	Environmental Scientis
Weather Conditions (clear, cloudy, rain, snow, etc	.):
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	rom thurst
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Signature: Herold J. Hanney	Date: 1/1/89
$\mathcal{D}$	Date:

# PRELIMINARY ASSESSMENT

# INFORMATION REPORTING FORM

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Indicate relative landmark locations (streets, buildings, streams, etc.).  Provide locations from which photos are taken.	
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### PRELIMINARY ASSESSMENT

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# PRELIMINARY ASSESSMENT INFORMATION REPORTING FORM

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hazardous wastes your installation handles. (See 40 CFR.)  X1. IGNITABLE (D001) (D002)	-aris 201.21 — 201.24.)	·	4. TOXIC
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RCRA	

# HAZARDOUS WASTE PERMIT APPLICATION

Consolidated Permits Program

I. EPA I.D. NUMBER
FN Y 000 2054351 745

RCRA		tion is required under	Section	300			F / 1/1/2	101012	017	1,5	21, 13		
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APPROVED (Vr. mo. & d	(4×)					OMMENTS							
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II. FIRST OR REVISED A		Control of the Contro				<b>是是是</b>	1		Part Par	C1.5			
Place an "X" in the appropriate revised application. If this is yo EPA I.D. Number in Itam I abo	ve.	You amondy Killow you	ui iaciii	11. Y S I	ether EPA I.	this is the first a .D. Number, or	ipplication you if this is a revise	are submit ed applicat	iting for ion, enti	your fa	scility or a facility's		
A. FIRST APPLICATION	place an ''X'' below an	d provide the appropri	ate date	e)		<del></del>			·		<del></del>		
1. EXISTING FACILITY	TY (See instructions for Complete item bel	r definition of "existin ow.)	g" facil	lity.			2.NEW FA	CILITY (C					
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B. REVISED APPLICATIO	N (place an "X" belou	and complete Item I	above)				73 74 75 -	5 77 78	EXPEC	TED T	O BEGIN		
1. FACILITY HAS IN							2. FACILI	TY HAS A	RCRA	PERMI	<b>T</b>		
III. PROCESSES – CODES	AND DESIGN CAP	ACITIES	145 E.A	1242	9 444	g of the state of the state of	72						
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2. UNIT OF MEASURE -	For each amount enter	d in column B(1), ent	er the c	ode f	rom t	he list of unit m	estura codas ha	alour that d	lacarihaa	•ho	:		
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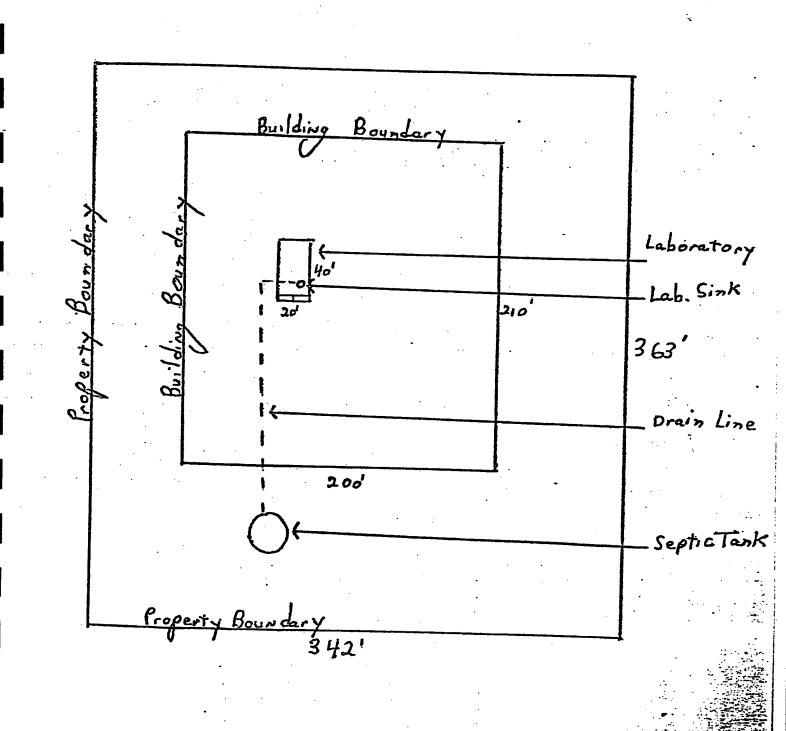
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EPA Form 3510-3 (6-80)

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Form 3510-3 (6-80)



REFERENCE NO. 4

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August 17, 1981

Permit Contact, Permits Adm. Branch Rm. 432 U.S. E.P.A. 26 Federal Plaza New York, NY 10007

Ref: Change of reporting status for Brinkmann Instruments, Inc., EPA I.D. # NYD002054351

Dear Sirs:

As per the attached form, Brinkmann Instruments submitted a Hazardous Waste Permit Application on November 19, 1980. Recent changes within our company indicate that we should withdraw or cancel this permit (Application).

On June 22, 1981, Brinkmann completely divested itself of its Clinical Diagnostics Division which was the sole source of potentially hazardous wastes being introduced into our local environment. Please note the enclosed announcement. From that time, all production and testing of these materials has been removed from the premises and Brinkmann Instruments is currently strictly an importer and distributor of scientific instruments and laboratory equipment.

We trust this information will permit your facility to delete our firm from your files as potential violators of current E.P.A. regulations. Should there be any change in our current status insofar as the utilization of hazardous materials, we shall most certainly keep you informed.

Very truly yours,

J. D. Brown Vice President

JDB:en

Enclosure

cc: R. Gabel, D, Canfield, J. Nordstrom

REFERENCE NO. 5



June 19, 1981

FIG : ENCLOSE YORK, A 7.10007

Dear Clinical Customer:

On June 22, 1981 Brinkmann Instruments sold its Clinical Division to a newly formed company, Biochemical Diagnostics, Inc. The principals of this company are Allan Panetz, Richard Gordon and Amir Farooqi all of whom were formerly employed by Brinkmann in its Clinical Division. Needless to say they are intimately familiar with the production and service requirements of the clinical test kits and products sold previously by Brinkmann.

The items involved are basically all of the Steroid Tests and Drug Screening products covered by catalog numbers beginning with "35". Sample concentrators will still be sold by Brinkmann direct. The address and telephone number of the new company is listed below and all orders should be sent to them. It is our belief as well as theirs that a smaller company devoted 100% to the manufacture and sale of these products can provide faster deliveries and response to clinical customer requirements. All of us at Brinkmann Instruments wish them much success in this new venture. Biochemical Diagnostics, Inc. will be ready to process customer orders within a day of the transfer of the company's assets on June 22nd. At the same time, we at Brinkmann thank you for your past orders for these particular products.

Very truly yours,

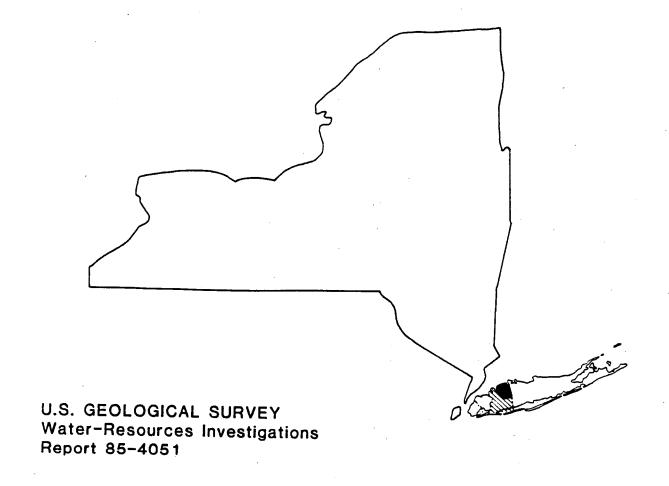
James D. Brown Vice President

> Biochemical Diagnostics, Inc. 121 Toledo Street Farmingdale, NY 11735

(516) 293-1206

Telex: - Att.Biochem 230199 SWIFT UR

Hydrogeology and Ground-Water Quality of the Northern Part of the Town of Oyster Bay, Nassau County, New York, in 1980



Prepared in cooperation with

NASSAU COUNTY DEPARTMENT OF PUBLIC WORKS



HYDROGEOLOGY AND GROUND-WATER QUALITY OF THE NORTHERN PART OF THE TOWN OF OYSTER BAY, NASSAU COUNTY, NEW YORK, IN 1980

By Chabot Kilburn and Richard K. Krulikas

U.S. GEOLOGICAL SURVEY

Water-Resources Investigations Report 85-4051

Prepared in cooperation with the
NASSAU COUNTY DEPARTMENT OF PUBLIC WORKS



Syosset, New York

# Hydrogeology and Ground-Water Quality of the Northern Part of the Town of Oyster Bay, Nassau County, New York, in 1980

by Chabot Kilburn and Richard K. Krulikas

#### **ABSTRACT**

This report presents hydrogeologic and water-quality data from the northern part of the Town of Oyster Bay, in the north-shore area of Long Island. The ground-water reservoir underlying the area consists of clay, silt, sand, and gravel layers that form six hydrogeologic units. The units are, from bottom to top, the Lloyd aquifer, Raritan clay, Magothy aquifer, Port Washington aquifer, Port Washington confining unit, and the upper glacial aquifer. Crystalline bedrock underlies the Lloyd aquifer and forms the base of the ground-water system.

The regional drought of 1962-67 caused ground-water levels to decline as much as 16 feet, but the water-table altitude in 1980 equaled or exceeded predrought levels. Water levels measured in wells screened in the lower part of the Magothy aquifer and in the Lloyd aquifer throughout much of the area are still below those measured before the drought but are recovering. Water levels in wells screened in the Lloyd aquifer along the north shore have been declining since the mid-1970's.

Ground water in some areas contains nitrates, volatile organic compounds, and chloride in concentrations that exceed New York State drinking-water standards. Contamination is limited largely to the upper glacial aquifer and upper part of the Magothy aquifer.

Saltwater has been reported in some wells along the shore but probably represents a natural condition rather than saltwater encroachment due to excessive pumping.

#### INTRODUCTION

Ground water is the sole source of drinking water for all of Nassau County. Because population and ground-water use have increased significantly since the 1950's, proper development of this resource requires detailed knowledge of the hydrogeologic environment and ground-water-quality. The U.S. Geological Survey began a study in cooperation with the Nassau County Department of Public Works to document the hydrogeology of the County. The area of this investigation is the part of the Town of Oyster Bay that lies north of Old County Road (fig. 1). The area contains approximately 71 mi<sup>2</sup>, or 63 percent of the town's 112-mi<sup>2</sup> area.

#### **HYDROGEOLOGY**

The ground-water reservoir underlying the northern part of the Town of Oyster Bay consists of unconsolidated glacial deposits of Pleistocene age and coastal-plain deposits of continental and marine origin of Late Cretaceous age. These unconsolidated deposits consist of gravel, sand, silt, and clay and are underlain by bedrock of early Paleozoic and (or) Precambrian age. The bedrock, which is relatively impermeable, forms the base of the ground-water reservoir.

The thickness, character, and water-bearing properties of the aquifer and the relationships between hydrogeologic and geologic units underlying the study area are depicted in table 1. The correlations should be considered direct relationships as implied in the tables. The upper and lower boundaries of the hydrogeologic units are determined mainly from gross lithologic differences between units rather the age of the deposits, which forms the basis for geologic correlations. For example, the upper and lower limits of the confining units (Port Washington confining unit and Raritan clay) are placed at intervals where the lithologic sequence changes from predominantly clay to sand or sand and gravel, and these positions may have no time-stratigraphic significance. For this reason, and because differentiation between sediments of Pleistocene and Cretaceous age is difficult and uncertain, it is possible that some deposits of Pleistocene age have been included in the upper part of the Magothy aquifer, which, by present definition, is approximately equivalent to the Magothy Formation-Matawan Group, undifferentiated, of Late Cretaceous age. The three hydrogeologic sections (pl. 1B) show the inferred extent, lateral and vertical relationships, and the variations in depth, thickness, lithology, and structure of these units.

#### **Description of Hydrogeologic Units**

#### Bedrock

Bedrock of early Paleozoic and(or) Precambrian age underlies all of western Long Island (Fisher and others, 1962). The bedrock generally consists of metamorphic and igneous crystalline rocks—schist, gneiss, and granite—and lies at depths ranging from about 350 ft below sea level along the north shore to about 950 ft below sea level in the southeast part of the study area (pl. 2A, and hydrogeologic sections, pl. 1B).

Bedrock is generally regarded as the base of the ground-water reservoir on Long Island because of its density and low permeability. No wells in the Town of Oyster Bay are known to obtain water from bedrock.

#### Lloyd Aquifer

The Lloyd aquifer is the equivalent of the Lloyd Sand Member of the Raritan Formation of Late Cretaceous age (Cohen and others, 1968, p. 18). It consists of discontinuous layers of gravel, sand, sandy clay, silt, and clay, and lies roughly parallel to the bedrock surface at depths ranging from about

System.	Series	Geologic unit	Hydrogeologic unit	Approximate range in thickness (feet)	Character of deposits forming geologic unit (modified from Swarzenski, 1963, and Isbister, 1966)	Water-bearing properties
QUATERNARY	Holocene	Undifferentiated artificial fill, salt- marsh and swamp deposits, stream alluvium, and shore deposits		0 - 50	Sand, gravel, silt, and clay; organic mud, peat, loam, and shells. Colors are gray, green, black, and brown.	Permeable zones near the shore and in stream valleys may yield small quantities of fresh or brackish water at shallow depths. Clay and silt beneath the north-shore harbors retard saltwater encroachment and confine underlying aquifers.
	Pleistocene	Upper Pleistocene deposits	Upper glacial aquifer	10 - 380	Till, composed of unsorted clay, sand, gravel, and boulders. Outwash deposits of stratified brown sand and gravel. May also contain some lacustrine and marine deposits consisting of clay, silt, and sand; locally fossiliferous.	Till, relatively impermeable, may cause local conditions of perched water and impede downward percolation of precipitation.  Outwash deposits of sand and gravel are highly permeable. Wells screened in glacial outwash deposits yield as much as 1,750 gal/min. Specific capacities of large-capacity wells range from 14 to 175 (gal/min)/ft of drawdown. Water is generally fresh and unconfined but may locally contain saltwater near shores.
CRETACEOUS - QUATERNARY	Cretaceous, Pleistocene, and Holocene(?)	Deposits of Holocene(?) and Pleistocene age, undifferentiated. May locally include eroded remnants of the clay member of the Raritan Formation.  Unconformity	Port Washington confining unit	0 - 360	Clay, solid and silty, gray, gray-green, white, red, mottled, and brown, containing lenses or layers of sand or sand and gravel. May locally contain lignite, shells, foraminifera, and other microfossils.	Relatively impermeable throughout much of the area. May be moderately to highly(?) permeable in areas adjacent to inferred limit of Magothy aquifer where sand and sand and gravel content may be large. Confines water in underlying Port Washington and Lloyd aquifers but does not prevent movement of water between upper glacial aquifer and Port Washington aquifer. Lenses of sand and sand and gravel provide sources of water supply and may permit local interchange of water with adjacent formations. One large-capacity well had a reported yield of 2,000 gal/min with a specific capacity of 43 (gal/min)/ft of drawdown. Coarser deposits may locally contain saltwater near shores.
	Upper	Deposits of Pleistocene age, undifferentiated, and (or) local erosional remnants of the Lloyd Sand Member of the Raritan Formation.	Port Washington aquifer	0 - 170	Sand, fine to coarse, white, yellow, gray, and brown, or gray and gravel with inter-bedded clay, silt, and sandy clay.	Moderately to highly(?) permeable. One large- capacity well had a reported yield of 1,100 gal/min with a specific capacity of 11 (gal/min)/ft of drawdown. Water is confined under artesian pressure. Generally contains freshwater but may have high iron content.

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CRETACEOUS	Upper Cretaceous	Matawan Group Magothy Formation- undifferentiated	Magothy aquifer	0 - 610	Clay, silt, sandy clay, and sand, fine to medium, clayey, white, gray, yellow, pink, and multicolored, in lenticular beds. May contain lenticular beds of coarse sand and gravel in lower part of unit. Lignite, pyrite, and iron oxide concretions may occur throughout the unit.	Moderately to highly permeable. Wells screened in lower part of aquifer yield as much as 1,400 gal/min. Specific capacities of large-capacity wells commonly range from 10 to 50 (gal/min)/ft of drawdown but may be as high as 80 (gal/min)/ft. Aquifer is principal source for public supply. Water is generally of excellent quality. Degree of confinement under artesian pressure is variable; however, artesian conditions increase with depth. Hydraulic continuity may exist between the Magothy aquifer and contiguous Pleistocene aquifers.
		Clay member	Raritan clay confining unit	0 - 185	Clay, solid and silty, gray, white, red, and mottled. May contain lenses or layers of fine to medium sand which may locally contain gravel. Sand layers frequently occur near top of unit. Lignite and pyrite are common.	Relatively impermeable. Confines water in underlying Lloyd aquifer but does not prevent movement of water between Magothy and Lloyd aquifers.
		Lloyd Sand Member Unconformity	Lloyd aquifer	0 - 195	Sand, fine to coarse, white, yellow, or gray, and gravel, commonly in a clayey matrix. Contains lenses and layers of solid or silty clay. Beds are usually lenticular and frequently show great lateral changes in composition.	Moderately permeable. Large-capacity wells may yield as much as 1,600 gal/min; specific capacities commonly range from 10 to 19 (gal/min)/ft of drawdown. Water is confined under artesian pressure; some wells flow. Water is generally of excellent quality but may have high iron content.
		Crystalline rocks	Bedrock	Not known	Metamorphic and igneous rocks; muscovite-biotite schist, gneiss, and granite(?). May have weathered zone at top.	Relatively impermeable. Contains some water in fractures, but impracticable to develop owing to low permeability.

200 ft below sea level along the north shore to about 700 ft below sea level in the southeast part of the study area (pl. 2B). Its thickness ranges from 0 to 250 ft from northwest to southeast, respectively.

The Lloyd aquifer is a major aquifer in the Town of Oyster Bay. It is probably hydraulically continuous with the adjacent Port Washington aquifer and upper glacial aquifer in the northern part of the study area. Water in the Lloyd aquifer is confined under artesian pressure beneath the Raritan clay.

Well yields during test pumping of large-capacity public-supply wells screened in the Lloyd aquifer have ranged from  $500~{\rm gal/min}$  to as much as  $1600~{\rm gal/min}$ .

#### Baritan Clay

The Raritan clay is a distinct hydrogeologic unit that extends throughout much of the Town of Oyster Bay (pl. 3A). In this area, the Raritan clay may be equivalent to the unnamed clay member of the Raritan Formation of Late Cretaceous age. The Raritan clay consists mainly of light to dark gray, red, white, or yellow clay and variable amounts of silt, and clayey silty fine sand. Sandy beds of varying thickness are common. The top of the Raritan clay is roughly parallel to that of the underlying Lloyd sand member. The upper-surface altitude of the Raritan clay ranges from 150 ft below sea level along the north shore to about 550 ft below sea level in the southeastern part of the study area. Its thickness ranges from 0 to 200 ft from northwest to southeast, respectively.

The Raritan clay is a significant hydrogeologic unit because it confines water in the underlying Lloyd aquifer. Although its hydraulic conductivity is very low, it does not entirely prevent movement of water between the Magothy and Lloyd aquifers. Some public-supply and other wells obtain part of their water supply from the sandy zones in the upper part of the Raritan clay.

### Magothy Aquifer

The Magothy aquifer is the equivalent of the Matawan Group-Magothy Formation undifferentiated of upper Cretaceous age. Deposits in this unit consist of beds and lenses of light-gray, fine to coarse sand with some interstitial clay. Detailed lithologic descriptions are given in Soren (1978); Ku and others (1975); and Jensen and Soren (1974).

The top of the Magothy aquifer is not planar, unlike the surfaces of the underlying units. The Magothy surface was deeply eroded during Tertiary time and probably was considerably eroded in Pleistocene time. The upper surface altitude of the Magothy ranges from as high as 200 ft above sea level in the center of the study area to 200 ft below sea level along the northeast edge of the study area (pl. 3B). Its thickness ranges from 0 to 650 ft from northwest to southeast, respectively.

The Magothy aquifer is the principal aquifer underlying Long Island and is the island's main source of water for public supply. The sand beds within the aquifer are moderately to highly permeable. The reported yields during

pumping tests of several public-supply wells screened in the Magothy aquifer in the Town of Oyster Bay ranged from 300 gal/min to as much as 1,500 gal/min. The average yield was about 1,000 gal/min.

The large amount of clay in the upper half of the aquifer causes the water to become increasingly confined with depth. Along the north shore, the Magothy aquifer is probably in hydraulic continuity with the adjacent Port Washington aquifer. The Magothy also has a generally high degree of hydraulic continuity with the overlying upper glacial aquifer, but the degree of continuity may vary considerably from place to place.

#### Fort Washington Aquifer

Two previously unrecognized hydrogeologic units in the northern part of the Town of Oyster Bay are defined as the Port Washington aquifer and Port Washington confining unit. The units were first recognized in the northern part of the Town of North Hempstead (Kilburn, 1979). The inferred limits of the units are shown in plates 4A and 4B, and their relationships to the other hydrologic units are shown on the hydrogeologic sections on plate IB.

The Port Washington aquifer is a sequence of deposits of Pleistocene and (or) Late Cretaceous age that underlie the north-shore area of the Town of Oyster Bay. The deposits form a distinct hydrogeologic unit that rests upon bedrock and is overlain by a thick sequence of confining clay. The south edge of the deposits overlap and abut the adjacent Cretaceous units. The sediments of the Port Washington aquifer form part of the valley fill in the channels cut into the Cretaceous deposits. These deposits consist largely of sand or sand and gravel and varying amounts of interbedded clay, silt, and sandy clay.

The altitude of the top of the Port Washington aquifer ranges from 150 ft below sea level along the north shore to 450 ft below sea level along the south shore (pl. 4A). Its thickness ranges from 0 to more than 150 ft in the central parts of the study area.

The Port Washington aquifer is moderately to highly permeable and is a major aquifer in the northern parts of the Town of Oyster Bay. The reported yields during pumping tests of public-supply wells screened in the aquifer range from 300 gal/min to 1,200 gal/min. Water in the aquifer is confined beneath the Port Washington confining unit. The hydrogeologic relationships between the Port Washington aquifer and the abutting Lloyd, Magothy, and upper glacial aquifers, as shown in the hydrogeologic sections on plate 1B, suggest that these deposits could be in lateral hydraulic continuity. Potentiometric studies of the head in the Lloyd aquifer made by Swarzenski (1963), Kimmel (1973), and Kilburn (1979) tend to verify a lateral hydraulic continuity between the Port Washington and Lloyd aquifers.

#### Port Washington Confining Unit

The Port Washington confining unit is a sequence of deposits of Pleistocene or Late Cretaceous to Holocene(?) age that locally underlies the north shore. The unit consists mainly of clay and silt, with scattered lenses

of sand or sand and gravel. (See Kilburn, 1979, for a more detailed description.) The deposits that form the Port Washington confining unit overlie the Port Washington aquifer or overlap the adjacent Cretaceous units and may form part of the valley fill that occupies channels cut into the other Cretaceous deposits. The unit may locally include or consist of erosional remnants of the clay member of the Raritan Formation.

The altitude of the top of the Port Washington confining unit ranges from 100 ft above sea level in the central part of the study area to 300 ft below sea level along the northeastern part (pl. 4B). Its thickness ranges from 0 to more than 150 ft in the central part of the study area.

#### Upper Glacial Aquifer

The upper glacial aquifer consists of deposits of late Pleistocene and Holocene age that overlie the Magothy aquifer and the Port Washington confining unit and locally abut against or overlie the Port Washington aquifer. The extent and relationships of these deposits to the adjacent hydrogeologic units are shown on plate 1B.

The upper deposits consist mainly of stratified beds of fine to coarse sand and of sand and gravel but also contain thin beds of silt and clay interbedded with coarse-grained material. The outwash that constitutes the bulk of the upper Pleistocene deposits is yellow and brown or, in some places, gray. (See Perlmutter, 1949, and Kilburn, 1979, for further descriptions.)

The upper glacial aquifer, which contains the water table in most of the area, transmits all recharge to the underlying aquifers. Precipitation filtering downward to the water table is the principal source of ground-water recharge. In the past, the upper glacial aquifer was tapped as a water supply by many public-supply wells. Because it has become contaminated by cesspool effluents, fertilizers, and other substances, however, its use for public supply has decreased. Wells tapping the aquifer are now used mainly to supply water for domestic use, irrigation, and commercial and industrial purposes.

The sand and gravel deposits in the upper glacial aquifer are highly permeable and yield large amounts of water to properly constructed wells. The yields of large-capacity public-supply wells screened in the aquifer have been reported to range from 400 gal/min to 1,400 gal/min.

The recent deposits of Holocene age along beaches, streams, swamps, and the bottoms of bays and lakes have not been differentiated from the upper glacial aquifer because they are too thin.

#### **Correlation of Units**

The differentiation between deposits of Pleistocene and Cretaceous age throughout most of the northern part of the Town of Oyster Bay is uncertain. On Long Island, the contact between Pleistocene and Cretaceous deposits is an erosional unconformity that is commonly marked by an abrupt lithologic and

mineralogic change. In most of the study area, this boundary has not been delineated with confidence because of the lack of reliable lithologic data (cores and cuttings) and the uncertainty of recognizing this boundary from well drillers' logs, which formed the basis for correlations made during this study. The tops of the Cretaceous deposits may be lower in altitude than indicated on plates 2B, 3A, and 3B because they seem lithologically similar to the overlying Pleistocene deposits in most of the drillers' logs.

The inferred limits of some of the hydrogeologic units in plates 2, 3, and 4 have been extrapolated into areas where little or no data are available. Where the inferred boundaries are questionable, they are so indicated by dashed lines. In the hydrogeologic sections (pl. 1B), extrapolated and questionable extensions of the contacts of some of the units are indicated by question marks.

The hydrogeologic correlations of wells used in constructing the sections and maps in this report are given in table 7 (at end of report) with the altitudes of the tops of the hydrogeologic units penetrated by wells in the northern part of the Town of Oyster Bay.

#### **GROUND-WATER HYDROLOGY**

#### **Pumpage**

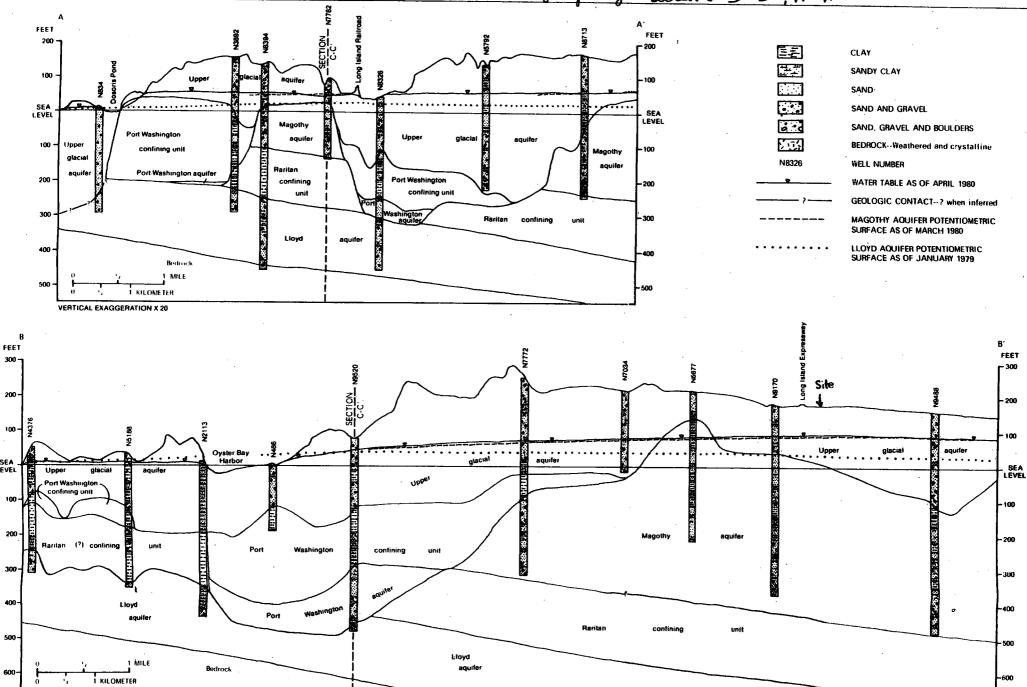
The total reported ground-water pumpage for all purposes in the northern part of the Town of Oyster Bay increased from 1.885 Bgal/yr (5.16 Mgal/d) in 1950 to 10.929 Bgal/yr (29.94 Mgal/d) in 1980. The total reported pumpage is plotted in figure 3A. During 1950-66, pumpage in the area increased at an average annual rate of about 291.6 Mgal/yr (798,900 gal/d); the graph does not reflect a significant effect from the 1962-67 drought. Beginning in 1967, however, the last year of the drought, the trend first reversed then leveled off until 1979. Pumpage during this time ranged between 7.95 Bgal/yr (21.8 Mgal/d) in 1967 and 9.94 Bgal/yr (27.2 Mgal/d) in 1971 and averaged 9.29 Bgal/yr (25.4 Mgal/d). A new upward trend may have started in 1980, when annual pumpage was reported to have been 10.92 billion gallons--an 11.6-percent increase over that in 1979. Whether this is a new trend or a short-term fluctuation will be ascertainable only from future records.

Ground water for public supply and nonpublic supply in the Town of Oyster Bay is derived principally from the upper glacial, Magothy, and Lloyd aquifers. The total amounts pumped from each of the aquifers and the Port Washington aquifer and confining unit are shown in figure 3B.

Ground-water pumpage from the Magothy aquifer is far larger than that from the other aquifers (fig. 3B) and, therefore, determines the trend of the graphs of total pumpage and public-supply withdrawals shown in figure 3A. Pumpage from the Magothy aquifer since 1976 (fig. 3B) has shown a significant upward trend that, through 1980, has increased by an average of more than 680 Mgal/yr (1.86 Mgal/d). This is largely because of a decline in public-supply withdrawals from the upper glacial aquifer and Port Washington confining unit (fig. 3C), but this loss is being made up by increasing pumpage from the

**VERTICAL EXAGGERATION X20** 

FEET



Hydrogeology and Grondwater Quelity of the northern Best of the Town of Gester Bay, Nassaw Conty, in 1986, EET

[6560-01]

#### (FRL 910-31

# AQUIFERS UNDERLYING NASSAU AND SUPFOLK COUNTIES, NEW YORK

#### Determination

Notice is hereby given that pursuant to Section 1424(e) of the Safe Drinking Water Act (42 U.S.C. 300f. 360h-3(e); 88 Stat. 1660 et seq.; Pub. L. 93-523) the Administrator of the Environmental Protection Agency has determined that the aquifer system underlying Nassau and Suffolk Counties, Long Island. New York, is the principal source of drinking water for these counties and that, if the aquifer system were contaminated, it would create a significant hazard to public health.

#### BACKGROUND

The Safe Drinking Water Act was enacted on December 16, 1974. Section 1424(e) of the Act states: "If the Administrator determines, on his own initiative or upon petition, that an area has an aquifer which is the sole of principal drinking water source for the area and which, if contaminated, would create a significant hazard to public health, he shall publish notice of that determination in the FEDERAL RECISIER. After the publication of any such notice, no commitment for Federai financial assistance (through a grant, contract, loan guarantee, or otherwise) may be entered into for any project which the Administrator determines may contaminate such

aquifer through a recharge zone so as to create a significant hazard to public health but a commitment for Federal financial assistance may, if authorized under another provision of law, be entered into to plan or design the project to assure that it will not so contaminate the aduler."

On January 21, 1975, the Environmental Defense Fund petitioned the Administrator to designate the aquifers underlying Nassau and Suffolk Counties. Long Island, New York, as a sole source aquifer under the provisions of the Act. A notice of receipt of this petition, together with a request for comments. was published in the FEDERAL REGISTER. Thursday, June 12, 1975. Written comments were submitted by the Environmental Defense Fund (EDF) on August 7, 1975, supporting their petition. A letter from the Director of the Nassau-Suffolk Regional Planning Board, dated October 1, 1976, requested that designation be delayed until after the completion of the areawide waste management (208) planning process for Long Island.

Because of the limited response to the FEDERAL REGISTER notice, EPA issued a press release and mailed an information sheet to elected officials and environmental groups on Long Island in March 1977. In addition, a presentation was made to the Citizens Advisory Committee (CAC) of the 208 planning agency and to the executive committee of the Long Island Water Conference. In response to these activities EPA received three comments: a letter from EDF questioning why project review would exclude direct Federal projects, a letter from a member of the East Hampton Planning Board expressing support for the designation, and a letter from the CAC requesting that designation be delayed until after the completion and approval of the Long Island 208 plan.

In considering the comments received, we could not agree with the letters requesting further delay since we do not believe that the review process under Section 1424(e) will constrain the options of 208 planning.

On the basis of the information which is available to this Agency, the Administrator has made the following findings, which are the basis for the determination noted above:

(1) The aquifers underlying Nassau and Suffolk Counties are the sole or principal drinking water source for the area. They supply good quality water for about 2.5 million people. Current water supply treatment practice for public supplies is generally limited to disinfection for drinking purposes, with some plants capable of nitrate removal. There are also numerous private sources. There is no alternative source of drinking water supply which could economically replace this aquifer system.

(2) The aquifer system is vulnerable to contamination through its recharge zone. Since contamination of a ground-water aquifer can be difficult or impossible to reverse, contamination of the the aquifer system underlying Nassau and Suffolk Counties. New York, would pose a significant hazard to those people dependent on the aquifer system for drinking purposes.

Among the determinations which the Administrator must make in connection with the designation of an area under Section 1424(e) is that the area's sole or principal source aquifer or aquifers, "if contaminated, would create a significant hazard to public health . . . Obviously, threats to the quality of the drinking water supply for such a large population could create a significant hazard to public health. The EPA does not construe this provision to require a determination that projects planned or likely to be constructed will in fact create such a hazard: it is sufficient to demonstrate that approximately 2.5 million people depend on the aquifer system underlying Nassau and Suffolk Counties as their principal source of drinking water, and that the squifer system is vulnerable to contamination through its recharge zone.

Section 1424(e) of the Act requires that a Federal agency may not commit funds to a project which may contaminate the aquifer system through a recharge zone so as to create a significant hazard to public health. The recharge zone is that area through which water enters into the aquifer system. Because of groundwater movement within these aquifers, the recharge zone is considered to be the entire area of Nassau and Suffolk Counties. However, both horizontal and vertical boundaries of the recharge zone are discussed in the background document under the section entitled "Area of Consideration."

The data upon which these findings are based are available to the public and may be inspected during normal business hours at the office of the Environmental Protection Agency, Region II. 26 Federal Plaza, New York, New York 10007. It includes a support document for designation of the aquifers underlying Nassau and Suffolk Counties, New York, and maps of the area within which projects will be subject to review.

A copy of the above documentation is also available at the U.S. Waterside Mall. Environmental Protection Agency. Public Information and Reference Unit. Room 2922, 401 M Street S.W., Washington, D.C. 20460.

The EPA has issued proposed regulations for the selective review of Federal financially assisted projects which may contaminate the aquifer system underlying Nassau and Suffolk Counties, New York, through the recharge

zone so as to create a significant hazard to public health. These proposed regulations were published in the FEDERAL REGISTER issue of September 29, 1977, and public comments were requested. They will be used as interim guidance for project review until their promulgation during 1978.

EPA. Region II. is working with the Federal agencies which may in the near future fund projects in the area of concern to EPA to develop interagency procedures whereby EPA will be notified of proposed commitments for projects which could contaminate the bicounty area's sole source aquifer system. Although the project review process cannot be delegated, the Regional Administrator in Region II will rely to the maximum extent possible upon any existing or future State and local control mechanisms in protecting the ground-water quality of the aquifer system underlying Nassau and Suffolk Counties, New York, Included in the review of any Federal financial. ly assisted project will be coordination with the State and local agencies. Their determinations will be given full consideration and the Federal review process will function so as to complement and support State and local mechanisms

Dated: June 12, 1978.

DOUGLAS M. COSTLE.
Administrator.
(FR Doc. 78-17067 Filed 6-20-78; 8:45 am)



# Uncontrolled Hazardous Waste Site Ranking System

A Users Manual (HW-10)

Originally Published in the July 16, 1982, Federal Register

United States
Environmental Protection
Agency

1984

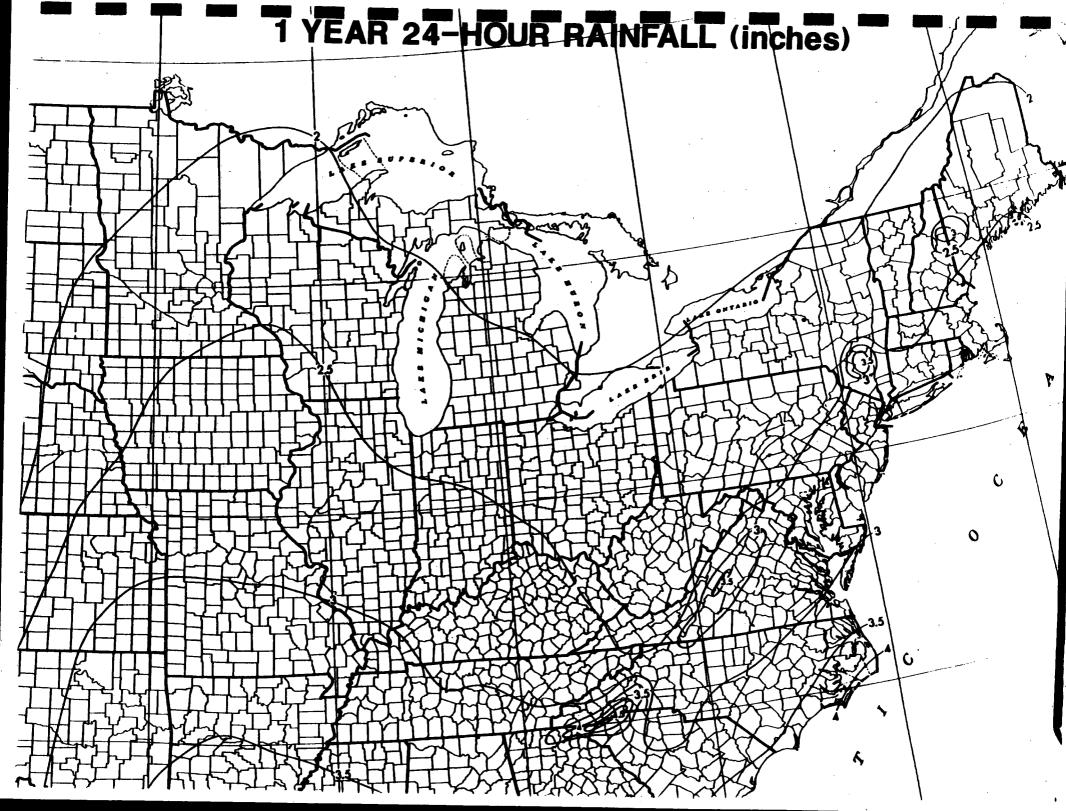
TABLE 2
PERMEABILITY OF GEOLOGIC MATERIALS\*

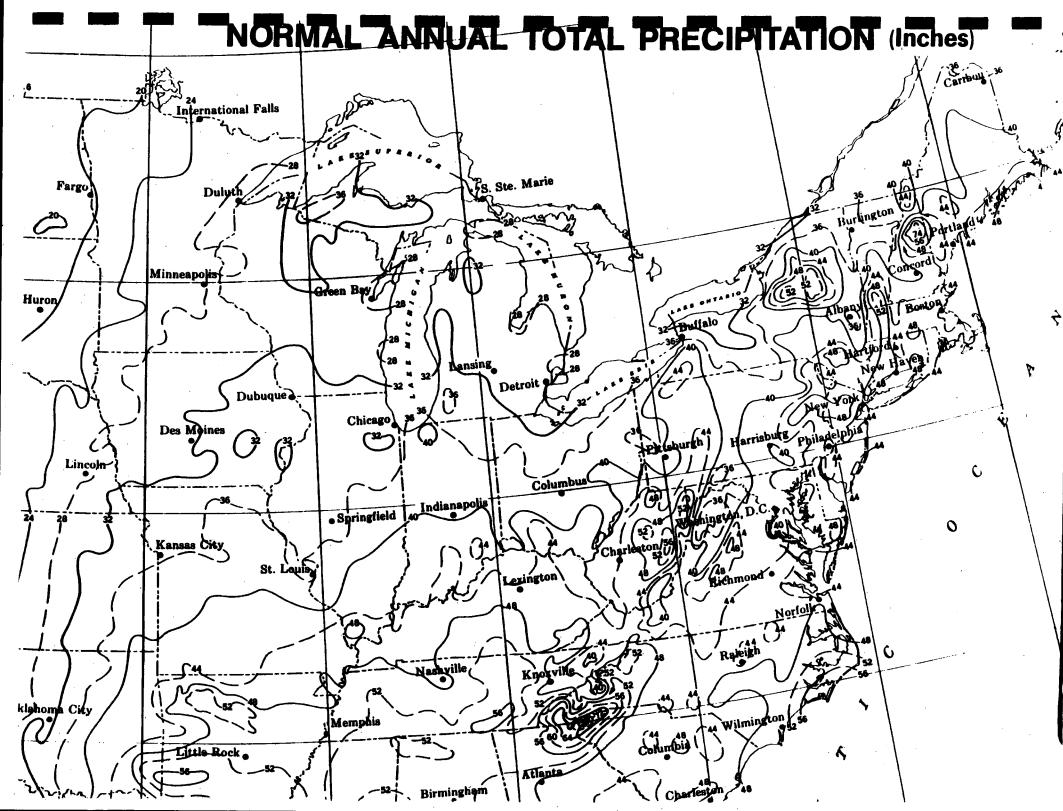
Type of Material	Approximate Range of Hydraulic Conductivity	Assigned Value
Clay, compact till, shale; unfractured metamorphic and igneous rocks	<10 <sup>-7</sup> cm/sec	0
Silt, loess, silty clays, silty loams, clay loams; less permeable limestone, dolomites, and sandstone; moderately permeable till	10 <sup>-5</sup> - 10 <sup>-7</sup> cm/sec	1
Fine sand and silty sand; sandy loams; loamy sands; moderately permeable limestone, dolomites, and sandstone (no karst); moderately fractured igneous and metamorphic rocks, some coarse till	10 <sup>-3</sup> - 10 <sup>-5</sup> cm/sec	2
Gravel, sand; highly fractured ligneous and metamorphic rocks; permeable basalt and lavas; tarst limestone and dolomite	>10 <sup>-3</sup> cm/sec	3

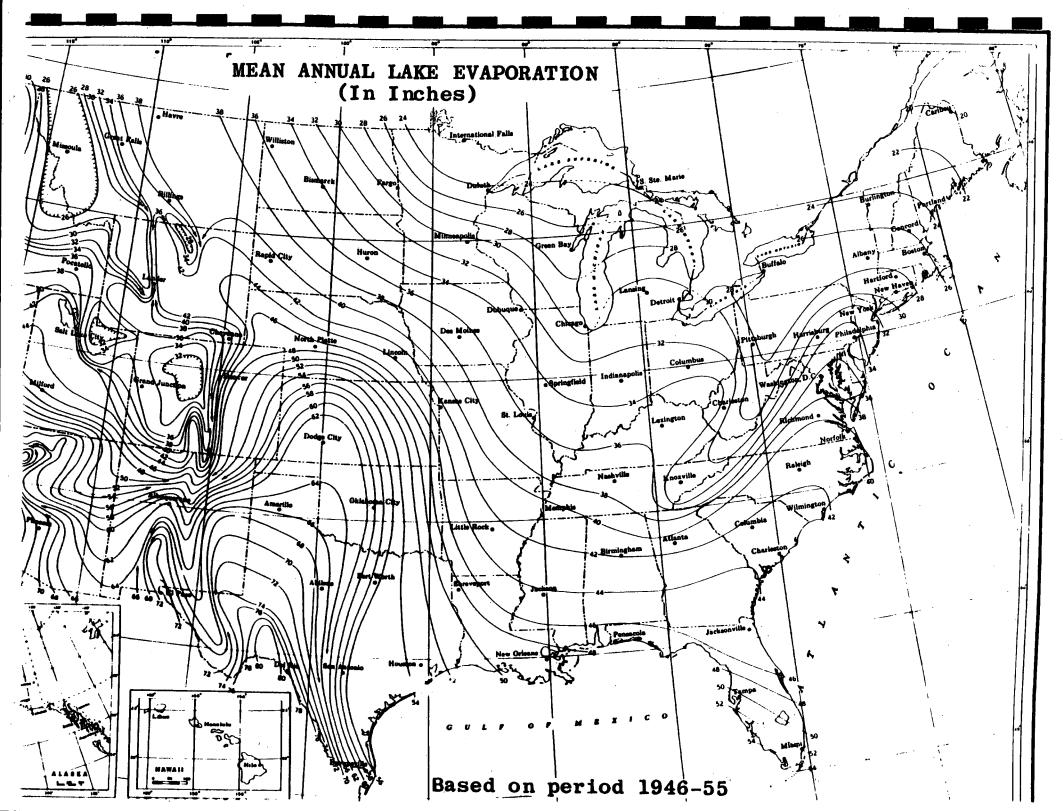
#### \*Derived from:

Davis, S. N., Porosity and Permeability of Natural Materials in Flow-Through Porous Media, R.J.M. DeWest ed., Academic Press, New York, 1969

Freeze, R.A. and J.A. Cherry, Groundwater, Prentice-Hall, Inc., New York, 1979







## HICKSVILLE WATER DISTRICT

4 DEAN STREET

HICKSVILLE, N. Y. 11802

PHONE 516) 931-0184

RESENTED

March 22, 1988

NUS Corporation
1090 King Georges Post Roady ( )
Suite 1103
Edison, New Jersey

Att: Mr. E.L. Leonard

Dear Mr. Leonard:

Enclosing please find the list you requested containing well numbers, depth and aquifers used by the nineteen (19) wells operated by the Hicksville Water District.

If I may be of any further assistance please feel free to contact me at the above number.

Very truly yours,

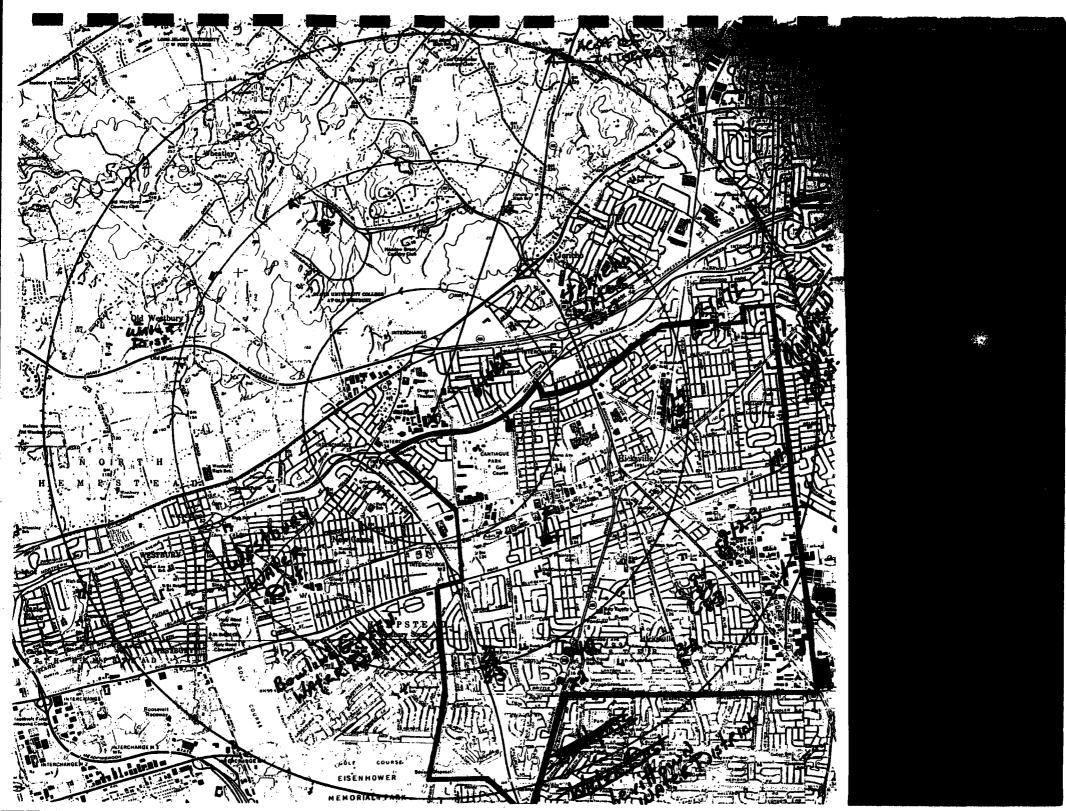
HICKSVILLE WATER DISTRICT

John J. McCrosson

Assistant Superintendent

Enc

JJM/jps



Willage of **Gld Wes**tbury

3021-6 32-5fe2-17

VILLAGE CLERK - TREASURER

R. BUSCARELLO

SUPT. PUBLIC WORKS

A. J. LINDON

VILLAGE JUSTICE W. F. RUEGER

MAYOR

GILBERT M. COLOMBO, JR.

TRUSTEES

E. A. SIMPSON

R. GACHOT

S. WEINSTEIN

H. BLAU

INCORPORATED MAY 10, 1924

1 STORE HILL ROAD

P.O. BOX 290

OLD WESTBURY, N.Y. 11568

RECEIVED

(516) 626-0800

APR 0.5 REC'D

HUS CORPURATION REGION II

SENT TO \_\_\_\_

April 4, 1988

Edward L. Leonard Nus Corporation 1090 King Georges Post Road Suite 1103 Edison, New Jersey 08837

Dear Mr. Leonard:

As requested, enclosed please find your area of interest map showing location and number of our wells and the out ine of our water district within the area.

The following is the ditional information on well depth and formation.

Well #1 N152 Depth of screen 478' Formation magothy

Well #4 N7549 Depth of screen 499' Formation magothy

Well #5 N8658 Depth of screen 610' Formation magothy

Village population served by these three wells is approximately 3200.

Blue lines on attached map indicates the boundary line of our water district within your area of interest.

> Arthur J. Lindon Supt. Public Works

AJL: hs Att.







160 Drexel Avenue Westbury, L.I., N.Y. 11590 516-333-0427

DONALD A. CROUCHLEY, Chairman FRANK J. IADEVIA, Secretary ALFRED ARDIS, Treasurer ITALO J. VACCHIO, Superintendent

REBEITED

March 21, 1988

NUS Corporation 1090 King Georges Post Road Suite 1103 Edison, New Jersey 08837

Attention: E. L. Leonard

Dear Mr. Leonard:

In response to your letter of March 17, 1988, (copy attached), please find below the requested information:

1. See attached map.

2.	STATE WELL NO.	WELL NO.	DEPTH	AQUIFERS
	N-101	6	341'	Magothy
	N-7785	7	400'	Magothy
	N-2602	9	805'	Lloyd
	N-5007	10	560'	Magothy
	N-5654	11	561'	Magothy
	N-5655	12	260'	Magothy
	N-6819	12A	270'	Magothy
	N-7353	14	390'	Magothy
	N-8007	15	564'	Magothy
	N-8497	16	544'	Magothy
	N-104510	17	600'	Magothy

- 4. Population estimated to be 24,000. All wells are interconnected.
- 5. None known.
- 6. l. Carle Place Water District
  - 2. Town of Hempstead Water District
  - 3. Inc. Village of Old Westbury
  - 4. Hicksville Water District
  - 5. Jericho Water District

Page #2.

Should you require any other information, please contact me at the above address.

Very truly yours,

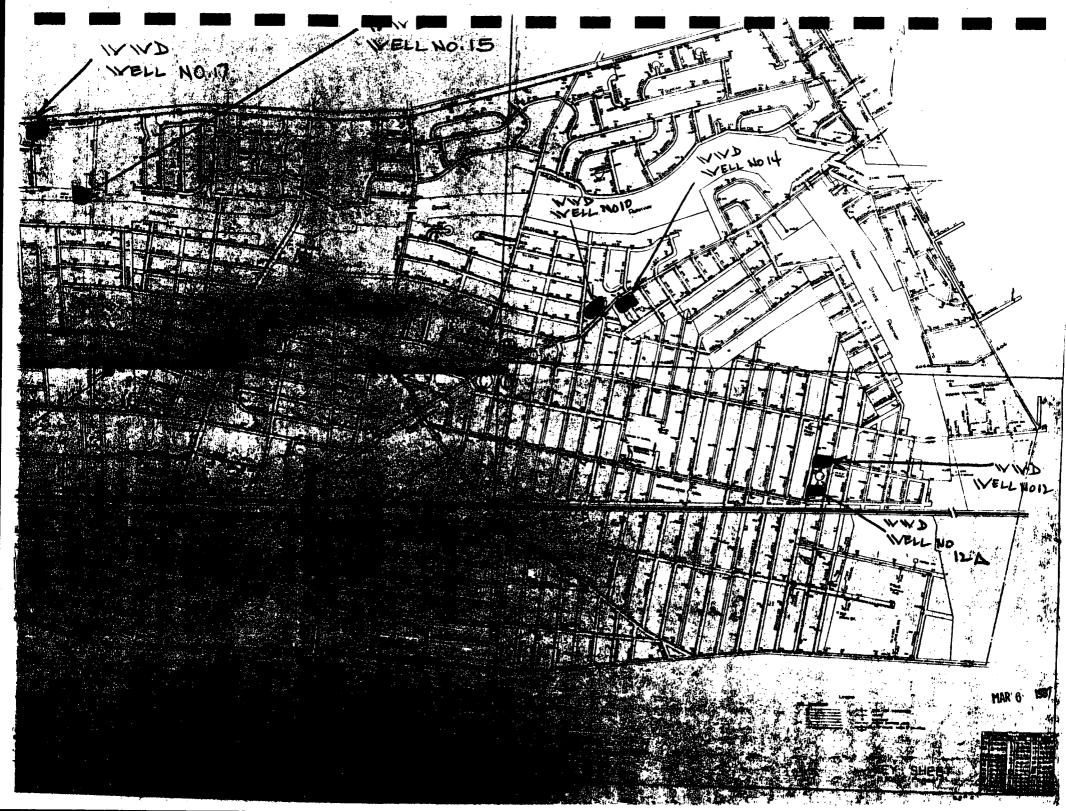
WESTBURY WATER DISTRICT

de Casolic

Italo J. Vacchio Superintendent

IJV/mh Enc.

3, 4





EDWARD P. BRACKEN, JR., CHAIRMAN NICHOLAS J. BARTILUCCI, TREASURER KENNETH J. DUNNE, SECRETARY

WILLIAM EVERS, SUPERINTENDENT GREG G. HENDRICKSON, OFFICE MANAGER 125 CONVENT ROAD SYOSSET, NEW YORK 11791 TEL. 921-8280

May 9, 1988

Mr. Edward Leonard NUS Corporation 1090 King Georges Post Road Suite 1103 Edison, New Jersey 08837

Dear Mr. Leonard:

Please find enclosed your "Marked-up" map which shows Jericho Water District boundaries in yellow. I have listed J.W.D. wells in red, with the number at each location. The wells are all interconnected with the rest of our wells, which total 20. We serve approximately 63,000 people.

The eight wells in this area are all in the magothy aquifer and their depths are as follows:

# 6	514'
# 7	484
#9.	565'
#10	453'
#14	615'
#15	535'
#16	490'
#22	459

Other water utilities have been marked on map with notation. If you have any further questions, please do not hesitate to call me.

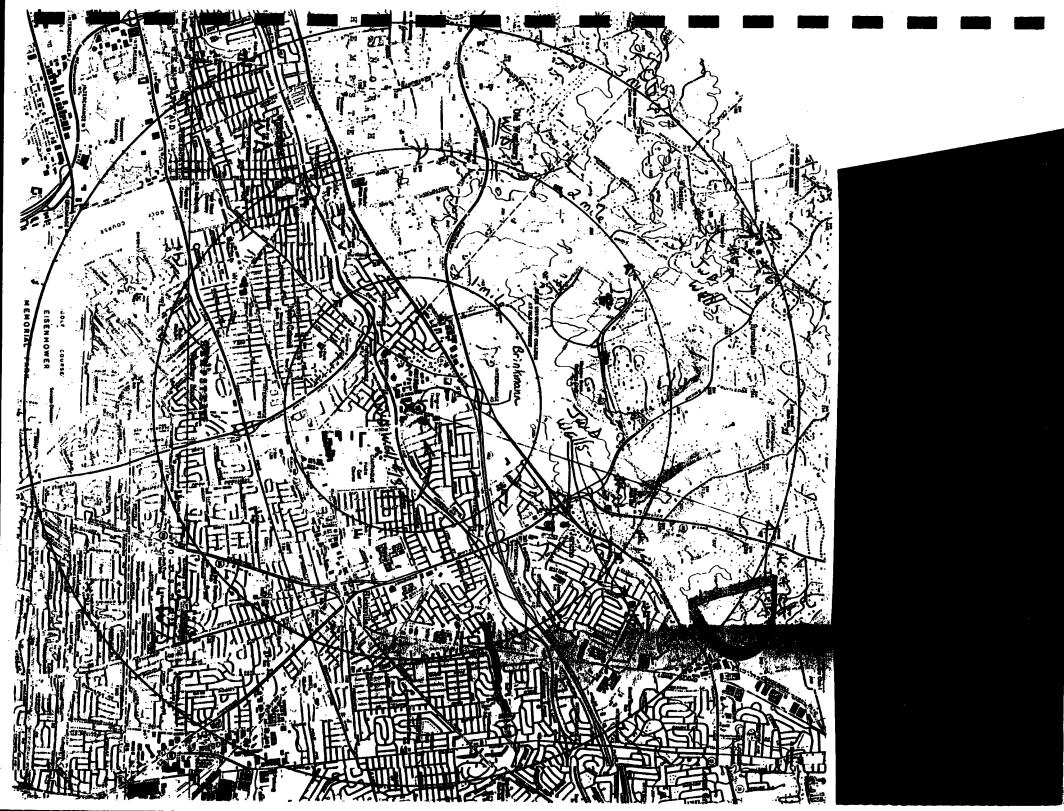
Very truly yours,

Board of Commissioners Jericho Water District

William Evers

District Superintendent

WE/ar



## Town of Hempstead

## Department

of

# Water

1995 PROSPECT AVENUE, EAST MEADOW, N.Y. 11554 (516) 794-8300



DANIEL DAVIS, P.E. Commissioner

ROBERT D. LIVINGSTON, JR. Receiver of Taxes

DANIEL M. FISHER, JR.

EUGENE L WEISBEIN
JOSEPH G. CAIRO, JR.
MARTIN B. BERNSTEIN
RICHARD V. GUARDINO, JR.
ANGIE M. CULLIN
PATRICK A. ZAGARINO

ELLO ERSON

April 6, 1988

Mr. Edward L. Leonard NUS Corporation 1090 King Georges Post Road Suite 1103 Edison, NJ 08837

Dear Mr. Leonard:

Below is the information requested in your letter of March 30, 1988. There are two wells located on Iris Place in the Bowling Green Estates Water District. Both wells are located in the magothy range and serves a population of approximately 12,000. Well #1, N-8956 is 535 feet deep; Well #2, N-8957 is 598 feet deep.

I am returning your map with the locations of the above wells marked in red.

If you should require any further assistance, please do not hesitate to contact me at (516) 794-8300, Ext. 204.

Very truly yours,

Harold V. Morgan

Administrative Assistant

HVM:th

enc. (2)

RECEIVED

FPR 0 8 RETS WUS CORPORATION REGION I: SENT TO





NEW YORK STATE DEPARTMENT OF HEALTH DIVISION OF ENVIRONMENTAL PROTECTION BUREAU OF PUBLIC WATER SUPPLY PROTECTION



## NASSAU COUNTY

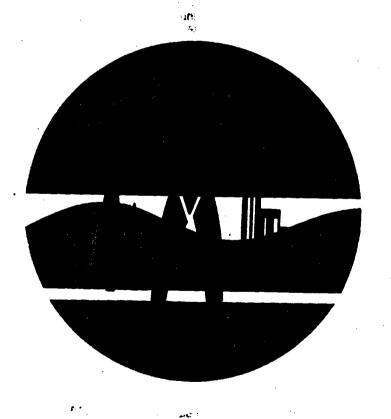
ID N	COMMUNITY WATER SYSTEM	POPULATION	SOURCE
Mun	icipal Community		
1	Albertson Water District	. 13500	. Weelis
ż	Bayville Village	7500	. Weell Is
3	Bayville Village	.32000	. We IIS
ŭ	Bowling Green Water District	. 12000	Weells
5	Carle Place Water District.	. 11000	Wells
. 6	Carle Place Water District Citizens Water Supply Company	30000	Wells
7	Deforest Drive Association	25	Wells
á	Deforest Drive Association, East Meadow Water District	52000	Wells
9	Farmingdale Village	7946	Wells
10	Franklin Square Water District	20000	We IIs
11	France Village	38272	Weile
12	Freeport Village	22506	Weells
13	Cardon City Village	22027	Welle
14	Garden City Village Glen Cove City	24618	Weelle
15	Homostood Village	10101	Lianile
<b>-</b> 16	Hempstead Village	59000	Westle -
- 10 17	lamaica Mater District	129000	West to
- 18	Jamaica water Supply Company	64000	Weste
	Jericho Water District Levittown Water District Lido-Point Lookout Water District.	- 64000	Wesile
19	Levittown water District	10000	West to
20	Ligo-Point Lookout water District.	. 10000	. West 1 S
21	Locust Valley Water District Long Beach City	20073	West 15
22	Long Beach City	.340/3	, WERTS
23	Long Island Water Corporation	270930	WENTS
24	Manhasset-Lakeville Water District	. 44/30	, WE-115
- 25	Massapequa Water District Mill Neck Estates Water Supply	.52000	.W62 ! 15
26	Mill Neck Estates water Supply	240	. WEETIS
27	Mineola Village	.20600	. Wee 115
28	New York Water Service Old Westbury Village Oyster Bay Water District Plainview Water District	.1/2180	, Wee: 115
29	Old Westbury Village	3100	. WET 15
30	Oyster Bay Water District	10227	. WEST 15
31	Plainview water District	. 40000	, wee: 115
32	Plandome Village	2010	. Weet 115
33	Port Washington Water District	. 35000	, Wee 115
34	Rockville Centre Village	. 25405	, WEH IS
35	Rooseveit Field Water District	1040	, WED 15
36	Roslyn Water District	.2/500	, Wesi I S
37	Plandome Village	. 3002	, WERLS
38			
39	Sel-Bra Acres Water Supply	80	. Weels
40	South Farmingdale Water District.	. 49900	, WC:115
41	Sel-Bra Acres Water Supply South Farmingdale Water District. Split Rock Water Supply	25	. Well Is
42	Uniondale Water District	. 25000	. WEEL 15
43	WAST MARKSTRAN-MARKSTRAND GARDEN		
	Water District	. 32000	, WEELIS
44	Westbury Water District	.20050	. WEE'IIS
45	Williston Park Village	8216	. WENTS
<b>a.</b>			
	-Municipal Community		
46	Community Hospital at Glen Cove	. 1350	, Well Is
47	Planting Fields Arboretum Stuart, Walker, Zimmer Water Suppl	90	, Well Is
48	Stuart, Walker, Zimmer Water Suppl	y 41	. Wells

# Catalog of Recharge Basins on

Long Island, New York,

in 1969

by
G. E. Seaburn and D. A. Aronson
U.S. Geological Survey



BULLETIN 70 1973

New York State Department of Environmental Conservation

#### RECHARGE BASINS

Most of the recharge basins on Long Island are unlined open pits that dispose of storm runoff from residential, commercial, and industrial areas, and from highways (Seaburn, 1970). About 30 basins solely dispose of effluent from sewage-treatment plants. Those basins were not included in the study and are not considered in this report. The area of basins that dispose of storm runoff generally ranges from 0.1 to 30 acres and averages 1.5 acres. The average depth below land surface is 10 feet, but the depth of a few is as much as 40 feet. Storm runoff to recharge basins flows by gutters to street inlets. The street inlets are interconnected by sewers that carry storm water into the basins, where it infiltrates moderately to highly permeable sand and gravel deposits above the ground-water reservoir. Figure 2 shows the location of all the inventoried recharge basins that received storm runoff on Long Island in 1969.

Design and construction of recharge basins on Long Island is regulated and approved by the local governments. Available information on completed basins varies greatly in detail. None of the departments of local government maintains an up-to-date catalog of the type of data presented here.

#### SOURCES OF DATA

Data were compiled from several sources. These included engineering drawings of proposed land developments; maps of topography, water-table contours, geology, and soils; and aerial photographs.

Most of the data were obtained from engineering drawings of land developments filed with the local government's department responsible for regulating and approving construction designs. 'In general, the drawings contain the following information: A basin's location, date of construction, design capacity, actual capacity, dimensions, use, bottom altitude, overflow altitude, and land-surface altitude.

U.S. Geological Survey topographic maps were used to verify basin locations and land-surface altitudes of many basins. Aerial photographs were also used to verify basin locations.

The recharge basins were plotted on base maps from which the nearest street intersection as well as the latitude and longitude of each basin were determined.

A water-table contour map developed by Kimmel (1971) was used to estimate water-table altitudes below each basin. Geologic maps (Fuller, 1914; Perlmutter and Geraghty, 1963; Swarzenski, 1963; Lubke, 1964; Isbister, 1966; and Soren, 1970) and soil maps (Lounsbury and others, 1928; and Warner, 1969) were used to determine geologic and soil environment at each basin.

### BRINKMANN INSTRUMENTS

at: 40°46'35"N Long: 73°33'12"W

TIS.	C	OI	Dataset: N	YHF.	Number	OI	kecords = 6	Group	= 1
REC	#	1	POP	1	HOUSE	1	DISTANCE	1 !	SECTOR
_	1	1	2187	' i	664	1	0.400000	1	1
	2	- 1	٥	1	0	i	0.810000	1	1
_	3	- 1	10416	ſ	3011	i	1.60000	, <b>f</b>	1
_	4	- 1	30600	1	9646	f	3. 20000	1	1
	5	- 1	62243	1	19143	ŧ	4.80000	1	1
	6	- 1	75346	1	21339	1	6.40000	l	1

	Pop.	House
1/4	2184	664
1/2	2187	664
1	12603	3675
2	43203	13321
3	105446	32464
4	180792	53803

# Protection of Environment

40

PARTS 190 TO 299 Revised as of July 1, 1988





### **Environmental Protection Agency**

(f) The commercial chemical products, manfacturing chemical intermediates, or off-specification commercial chemical products referred to in paragraphs (a) through (d) of this section, are identified as toxic wastes (T), unless otherwise designated and are subject to the small quantity generator exclusion defined in § 261.5 (a) and (g).

[Comment: For the convenience of the regulated community, the primary hazardous properties of these materials have been indicated by the letters T (Toxicity), R (Reactivity), I (Ignitability) and C (Corrosivity). Absence of a letter indicates that the compound is only listed for toxicity.]

These wastes and their corresponding EPA Hazardous Waste Numbers are:

		OI C.
Haz- ardous waste No.	Chemical abstracts No	Substance
U001	75-07-0	Acetaldehyde (i)
U034	75-87-6	
U187	62-44-2	
U005	53-98-3	
U240	194-75-7	1 - managing of 14 of 1-1960 GIAE-Ala
U112	141-78-6	Linesia caral (#11 (#2011/01/01/01/01/01/01/01/01/01/01/01/01/
U144	301-04-2	Acetic acid, lead(2+) set
U214	563-68-8	
500	93-76-5	Acetic acid, (2,4,5-trichlorophenoxy)-
F027	}	
U002	67-64-1	
U003	75-05-8	i i i i i i i i i i i i i i i i i i i
U004	98-86-2	
U005	53-96-3	1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -
U006	75-36-5	Limital's assessment (wheelt)
U007 U008	79-06-1	f i and the firm
U009	79-10-7	1 - 7 V/
U011	107-13-1 61-82-5	1
U012	62-53-3	1
U138	75-60-5	
U014	492-80-8	1 i a anima mandi Mariadi Mari
U015	115-02-6	
U010	50-07-7	
		1,1a,2,8,8a,8b-hexahydro-8a-methoxy-5-methyl-, [1aS-(1asipha, 8beta,8asipha,8balpha,8b
U157	56-49-5	Benz[j]aceanthrytene, 1,2-dihydro-3-methyl-
U016	225-51-4	Benz(c)acridine
U017	98-87-3	Bertzal chloride
U192	23950-58-5	I an amount alough which the first the Self Child Mark
U018 U094	58-55-3	Benziajanthracene
U012	57-97-6 62-53-3	An arminimental 1'15-militing
U014	492-80-8	Benzenamine (I,T)
U049	3185-93-3	Benzenamine, 4,4'-carbonimidoyibis[N,N-dimethyl-
U093	60-11-7	Berzenamine, 4-chloro-2-methyl-, hydrochloride
U328	95-63-4	Benzenamine, N.N-dimethyl-4-(phenylazo)- Benzenamine, 2-methyl-
U353	106-49-0	Benzonamine, 4-methyl-
U158	101-14-4	Benzenamine, 4,4'-methylenebis(2-chloro-
U222	636-21-5	Benzenamine, 2-methyl-, hydrochloride
U181	99-55-8	Benzenamine, 2-methyl-5-nitro-
U019	71-43-2	Benzene (i,T)
U038	510156	Berzeneacetic acid, 4-chloro-alpha-(4-chlorophenyl)-alpha-hydroxy-, ethyl ester
U030 U035	101-55-3	Constraint (-CLOHIO-4-DURIDIA-
U037	305-03-3 108-90-7	Benzenebutanoic acid, 4-{bis(2-chloroethyl)amino}-
U221	25376-45-8	Benzane, chloro-
J028	117-81-7	Benzanediamine, ar-methyl-
J069	84-74-2	1.2-Benzenedicarboxylic acid, bis(2-ethylhexyl) ester 1.2-Benzenedicarboxylic acid, dibutyl ester
J088	84-66-2	1,2-Benzenedicarboxytic acid, diethyl ester
J102	131-11-3	1,2-Benzenedicarbonyric acid, dimethyl ester
J107	117-84-0	1.2-Berzenedicarboxytic acid, dioctyl ester
J070		Benzene, 1,2-dichloro-
J071		Benzene, 1,3-dichloro-
J072 J060	106-46-7	Benzene, 1,4-dichloro-
WOU	72-54-8	Benzene, 1,1'-(2,2-dichloroethylidene)bis[4-chloro-

		40 CFR Ch. I (7-1-88 Edition)
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Wa:		No. Substance
_ No	2.	
	_	
UOT	. ,	
U22:		2-3   Benzene 1 3-dispoyanote-market en -
U239		0-7 Benzene, dimethyl- (I,T)
U201		6-3 1.3-Benzenediol
U127		4-1 Benzene, hexachioro-
U056	110-8	8-7 Benzene, hexahydro- (I)
U220	108-8	
U105		
U106	606-20	
U055		
U169	96-95	-3 Benzene, nitro-
U183	606-93	
U185	82-68	Benzene, pentachioronitro
U020	96-09	
U020	98-09	
U207	95-94	
U061	50-29	
U247	72-43	
U023	96-07-	- I worked by 111 46.6.6.000 Commission No F.
U234	99-35-	
U021	92-67-	.
U202	81-07-	
U203	94-58-	
U141	120-58-	
Liogo	84-54	1 1,3-MORPOCHESSICIE S./S. CORRECTED
U064	100-55-	1,3-bermodomole, 5-graphi-
U246		
-2-10	101-01-	21-1-Bertzopyran-2-one, 4-hydrony-3-(3-own-1-phend had 0
U022	50.00	2 2H-1-Benzopyran-2-one, 4-hydroxy-3-(3-oxo-1-phenyl-bueyl)-, & selts, when present at concentrations
U197	50-32-	, i parimoralibitati
U023	106-51-	
U085	98-07-7	
U021	1464-63-6	1 2.2 - Bittistene
U073	92-97-6	
U091	91-84-1	[1,1'-Bishery]-4.4'-Garring 9 of Assessment 1
U095	119-80-4	1 11 TOURS IN I A Chicago 2 21 America
	119-83-7	1 111 Topingrift 4.4 decise 3 2 topingrift
U225	75-25-2	) bromblothi
U030	101-55-3	
U128	87-66-3	113-6886ene 11234 s.h
U172	924-14-3	1 1 CARTAGORIA MANAGAM AND
U031	71-36-3	( i-sumpo (i)
U159	76-83-3	2-Butanana (I,T)
U160	1338-23-4	2-Butanone, percuide (R,T)
U963	4170-30-3	2-butting
U074	784-41-0	2-Butane, 1,4-dichtoro- (I,T)
U143	303-34-4	2-Bullencic acid, 2-methol, 7-fre a communication of the communication o
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	_	[15-[1alpha(Z),7(25°,391°),7asipha]]-
U031	71-36-3	TOTAL COLO (I)
U136	75-60-6	Caccelylic acid
U238	13765-19-0	Calcium chromate
	51-79-8	Carbarnic acid, ethyl ester
U178 U097	012-03-5	Carbanic acid methylairess
11000		
U114	111-54-8	Certamographic scirl 1 2-minutes in the
U062	1	CONTROL OF CONTROL
U215	2303-16-4	Carbamothioic acid, bis/1-mathetathol. 5/2/2 dishlared
U033	252-73-8	Carbamothicic acid, bis(1-methylethyl)-, S-(2,3-dichloro-2-propenyl) ester  Carbonic acid, dthallium(1+) set
U156	353-50-4   1	Certhonic differentials
U033	79-22-1 (	Arthonochloride and enabled and a second
U211		
U211	30-53-3 (	APDON Mitrachtwise
U035	75-87-6   (	Prioret
U035	305-03-3	Morembucil
	57-74-0	Mordano, aloha & garrena incomo
U026	101-00-110	morniphizh
U037	108-90-7	Norobenzene
U038	510-15-6 C	Morobenzilate
U039	59-50-7 lp	Chloro-m-cresol
		J

Environn

Haz-ardous waste No. U042 U044 U046 U047 U048 U049 U050 U051 U052 U055 U055 U055 U197 U056 U129 U057
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U108 **20**8 5 14 1 1 16 32 1 1 

Haz- ardous waste No.	Chemical abstracts No.	Substance
	110-75-8	2-Chloroethyl vinyl ether
U042 U044	67-66-3	Chloroform
U046	107-30-2	Chloromethyl methyl ether
U047	91-58-7	beta-Chloronaphthalene
U048	95-57-8	o-Chlorophenol 4-Chloro-o-toluidine, hydrochloride
U049	3165-93-3 13765-19-0	Chromic acid H <sub>2</sub> CrO <sub>4</sub> , calcium salt
U032 U050	218-01-9	Chrysene
U051	210 01	Creosote
U052	1319-77-3	Cresol (Cresylic acid)
U053	4170-30-3	Crotonaldehyde
U055	98-82-8 506-68-3	Curriene (i) Cyanogen bromide (CN)Br
U246	108-51-4	2.5-Cyclohexadiene-1,4-dione
U197 U056	110-82-7	Ovicherane (I)
U129	58 89 9	Cyclohexane, 1,2,3,4,5,6-hexachloro-, (1alpha,2alpha,3beta,4alpha,5alpha,6beta)-
U057	108-94-1	Cyclohexanone (I) 1,3-Cyclopentadiene, 1,2,3,4,5,5-hexachloro-
U130	77-47-4 50-18-0	Cyclophosphamide
U058 U240	194-75-7	2.4-D, saits & esters
U059	20830-81-3	Daunomycin
U060	72-54-8	DOD
U061	50-29-3	COT
U062	2303-16-4	Dialiate   Dibenz(a,h)anthracene
U063	53-70-3 189-55-9	Dibenzo[a,i]pyrene
U064	96-12-8	
U066 U069	84-74-2	
U070	95-50-1	o-Dichlorobenzene
U071	541-73-1	
U072	108-46-7	
U073	91-94-1 784-41-0	1 The second second at 1 The
U074	75-71-6	1 2
U078	75-35-	1.1-Dichlorosthylene
U079	156-60-9	
U025	111-44-	
U027	108-60-	T. C.
U024	111-91-	The same and the same
U081 U082	87-65-	2,6-Dictriorophenol
U084	542-75-	i 1,3-Dichloropropene
U065		
U108	44	
U028		N N-Digitivihydrazine
U086 U087		I a a a a a a a a a a a a a a a a a a a
U088		2 Diethyl phthalate
U089	56-53-	
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U092		2 p.Dimetrylaminoszoberzene
U094	, , , , , , , ,	
U098	119-93-	
U096		
UÖ9		
U091		8 1,2-Dimethylthydrazine
U10	1	9 2.4-Dimethylphenol
U10	2 131-11	
U10		
U10		
Ü10 U10	792 27	Di-n-octyl phthelete
U10	, , , , , , , , , , , , , , , , , , , ,	-1 { 1.4-Dioxane
U10	9 122-66	-7   1,2-Dighenyihydrazine
Ü11	0 142-84	
U11		

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Haz- ardous waste No.	Chernical abstracts No.	Substance
U041		Epichlorohydrin
U001 U174		Ethanal (I) Ethanamina, N-athyl-N-nitroso-
U155		i 1.2-Ethenediemine, N,N-dimethyl-N'-2-pyridinyl-N'-(2-thienylmethyl)-
U067	106-83-4	Ethane, 1,2-dibromo-
U076 U077		Ethane, 1,1-dichloro-
U131	1	Ethane, 1,2-dichloro-
U024		Ethene, 1,1'-[methylenebis(oxy)]bis[2-chloro-
U117		Ethane, 1,1-oxytis-(I)
U025 U184		Ethans, 1,1'-oxybis[2-chloro-
U208		Ethane, 1,1,1,2-letrachioro-
U209		Ethane, 1,1,2,2-letrachioro-
U218 U226		: Ethanethioamide : Ethane, 1.1.1-trichloro-
U227		Ethane, 1,1,2-trichloro-
U359		Ethanol, 2-ethoxy-
U173 U004		Ethanol, 2,2'-(nitrosoimino)bis-
U043		: Ethanone, 1-phenyl- : Ethane, chloro-
U042		Ethene, (2-chloroethoxy)-
U078		Ethene, 1,1-dichloro-
U079 U210		Ethene, 1,2-dichloro-, (E)- Ethene, tetrachloro-
U228		Ethene, trichloro-
U112		Ethyl acetate (I)
U113 U238		Ethyl acrylate (I)
U117		Ethyl other (I)
U114		! Ethylenebiadithiocarbamic acid, salts & esters
U067		Ethylene dibromide
U077 U359		! Ethylene dichloride ! Ethylene glycol monoethyl ether
U115		Ethylene oxide (I,T)
U116		Ethylenethicures
U076 U118		Ethylidene dichloride
U119		Ethyl methanesulfonate
U120		Fluoranthene
U122 U123		1 Formaldehyde 1 Formic acid (C,T)
U124	110-00-8	
U125		2-Furancarboxaldehyde (I)
U147 U213		2,5-Furancione   Furan, tetrahydro-(I)
U125		Furtural (I)
U124		Furturan (I)
U206 U206	18883 <del>-66-4</del>	Glucopyranose, 2-decxy-2-(3-methyl-3-nitrosoureido)-, D- : D-Glucose, 2-decxy-2-[[(methylnitrosournino)- carbonyl]amino}-
U126 U163		Glycidylaidehyde
U103		I Guanidine, N-methyl-N'-nitro-N-nitroso- I Hexachtorobenzene
U128		Hexachlorobutadiene
U130 U131	7.7	Hexachlorocyclopentadiene   Hexachloroethane
U132		Hexachiorophene
U243	1886-71-7	Hexachioropropene
U133 U086		Hydrazine (R,T)
U098		Hydrazine, 1,2-diethyl- Hydrazine, 1,1-dimethyl-
U099	540-73-8	Hydrazine, 1,2-dimethyl-
U109		Hydrazine, 1,2-diphenyi-
U134 U134		Hydrofluoric acid (C,T) Hydrogen fluoride (C,T)
U135		Hydrogen sulfide
U135	7783-06-4	Hydrogen sulfide H <sub>s</sub> S
U096 U116		Hydroperoxide, 1-methyl-1-phenylethyl- (R) 2-imidezolidinethione

Environmente

U217

Substance

Indeno(1,2,3-cd)pyrene

1,3-isobenzoturandione

Lead, bis(acetato-O)tetrahydroxytri-

Isobutyl alcohol (I,T)

Iron dextran

Lasiocarpine

Lead acetate

Lead phosphate

Lead subacetate

Maleic anhydride

Maleic hydrazide

Methane, bromo-

Methane, dibromo-

Methane, dichloro-

Methane, tribromo-

Methane, trichloro-

Methanol (I)

Methoxychlor

Methyl alcohol (I)

1-Methytbutadiene (I)

Methyl chloride (I.T)

Methyl chloroform

Methylene bromide

Methylene chloride

Methyl chlorocarbonate (I,T)

3-Methylcholanthrene 4,4'-Methylenebis(2-chloroaniline)

Methyl ethyl ketone (MEK) (I,T)

Methyl iodide ... Methyl isobutyl ketone (I) Methyl methacrylate (I,T)

4-Methyl-2-pentanone (I) Methylthiouracii

1-Naphthalenamine 2-Naphthalenamine

Naphthalene, 2-chloro-

1,4-Naphthalenedione

1,4-Naphthocuinone

alpha-Naphthylamine beta-Naphthylamine

10102-45-1 Nitric acid, thallium(1+) salt

Nachthalene

Methyl ethyl ketone peroxide (R,T)

Naphthalenamine, N,N'-bis(2-chloroethyl)-

2.7-Nachthalenediaulfonic acid, 3,3'-[(3,3'-

dimetry([1,1'-biphenyl]-4,4'-diyl)bis(azo)bis(5-4

Methyl bromide

Methane, trichlorofluoro-

Methane, iodo-

Methane, chloro- (I, T)

Methane, chloromethoxy

Methane, dichlorodifluoro-

Methanesulfonic acid, ethyl ester Methane, tetrachloro-Methanethiol (I, T)

4,7-Methano-1H-indene, 1,2,4,5,6,7,8,8-octachioro-2,3,3a,4,7,7a-hexahydro-

Mitomycin C 5,12-Naphthacenedione, 8-scetyl-10-((3-amino-2,3,6-trideoxy)-alpha-L-lyxo-her 7,8,9,10-tetrahydro-6,8,11-trihydroxy-1-methoxy-, (6S-cis)-

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Methapyrilene
1,3,4-Metheno-2H-cyclobuta(cd)pentalen-2-one, 1,1a,3,3a,4,5,5,5a,5b,6-decachloroctahydro

Methacrylonitrile (I, T)
Methanamine, N-methyl- (I)

Malononitrile

Melphalan

Mercury

Kepone

Lindane

MNNG

Chemical abstracts No.

193-39-5

9004-66-4 65-44-9 78-83-1 120-58-1

143-50-0 303-34-4

301-04-2

1335-32-8

7446-27-7 1335-32-6

58-89-9 70-25-7

108-31-6 123-33-1 109-77-3

148-82-3 7439-97-6 126-98-7 124-40-3 74-83-9

74-87-3

107-30-2 74-95-3

75-09-2 75-71-8

74-88-4 62-50-0 56-23-5

74-93-1 75-25-2

67-66-3 75-69-4

57-74-9

67-56-1

91-80-5 143-50-0

72-43-5 67-56-1

74-83-9 504-60-9 74-87-3

79-22-1

71-55-6

71-55-6 58-49-5 101-14-4 74-95-3 75-09-2

78-93-3 1338-23-4 74-88-4

108-10-1

80-62-6 108-10-1

58-04-2 50-07-7

134-32-7

494-03-1

91-59-8

91-20-3 91-58-7

130-15-4

130-15-4

134-32-7

91-59-8

72-57-1

20830-81-3

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U137

U139

U190

U140 U141

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U151

U152 U092

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U155 U142

**U247** 

**U154** 

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**U164** 

U010

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Hazardous waste No.

U008 U152 U008 U113 U118 U162 U194 U069 U148 U198 U191 U237

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erdou waste No.			Substance
U169	98-95-	3 Nitrobenzene (LT)	
Ü170	100-02-		
U171	79-46-	- I objection (P1)	·
U172	924-16-		
U173 U174	1116-54- 55-18-		
U176	759-73-		
U177	684-93-		
U178	615-53-2	N-Nitroso-N-methylurethane	•
U179 U180	100-75-4	N-Nitrosopiperidine	
U181	930-55-2	N-Nitrosopyrrolidine 5-Nitro-o-toluidine	
U193	1120-71-	1,2-Oxathiolans, 2,2-dioxide	•
U058	50-18-0	24-1,3,2-Oxezaonosonom-2-amine	
U115	75-21-8	N.N-bis(2-chloroethyl)tetrahydro-, 2-oxide Oxirane (I,T)	
U126	765-34-4		
U041	106-89-8		
U182	123-63-7	Paraidehyde	
U183 U184	608-93-5	6	
U185	76-01-7 82-68-8		
See	87-86-5	S CONTRACTOR CONTRACTOR (PCDAM)	
F027	1		•
U161	108-10-1	Pentanol, 4-methyl-	•
U186 U187	504-60-9	1.3-Pentadiene (I)	•
U188	62-44-2 108-95-2	Phenecetin Phenol	•
U048	95-57-8	Phenol. 2-chloro-	
U039	<del>59-5</del> 0-7	Phenol, 4-chloro-3-methyl-	•
U081	120-83-2	Phenol, 2,4-dichloro-	
U082 U089	87-85-0 56-53-1		
U101	105-67-0	Phenol, 4.4'-(1,2-disthyl-1,2-ethenediyl)bis-, (E)- Phenol, 2.4-dimethyl-	
U052	1319-77-3	Phenoi, methyl-	
U132	70-30-4	Phenol, 2.2'-methylenebis[3,4,6-trichlorp-	•
U170 See	100-02-7 67-88-5	Phenol, 4-nitro-	•
F027	e/-00-0	Phenol, pertachloro-	
See F027	58-90-2	Phonol, 2,3,4,6-tetrachloro-	
See F027	95-95-4	Phenol, 2,4,5-trichloro-	
See F027	88-08-2	Phonol, 2,4,6-trichloro-	
U150	148-82-3	L-Phenytalenine, 4-[bis(2-chloroethyl)amino]-	•
U145 U087	/	PROGRADUL SCILL SOCIONAL SOCIALISTS	
U189		Phosphorodithioic acid. O.O-diethyl S-methyl ests Phosphorus sulfide (R)	<b>*</b>
U190		Prihalic anhydride	•
U191	109-06-8	2-Picoline	•
U179 U192	100-75-4   23950-68-5	Piperidine, 1-nitroso-	
U194		Pronamide 1-Propanamine (I,T)	
U111	621-64-7	Propanamine, N-nitroso-N-propyl-	•
U110	145-04-1	Properamine, N-propert, (i)	•
U066 U063	96-12-8   1 78-87-5   1	Propens, 1.2-dibromo-3-chioro-	
U149		ropene, 1,2-dichloro- ropenedinistie	
J171	79-46-0	ropana, 2-nitro- (I,T)	•
1027	108-60-1	ropene, 2,2'-oxybis(2-chioro-	*
7193	1120-71-4	,3-Propene sultone ropenoic ecid, 2-(2,4,5-trichlorophenoxy)-	
300		- Arthur and Arthur an	
F027			
F027 J235	128-72-7 1	Propanol, 2,3-dibromo-, phosphate (3:1)	
F027 /235 /140	128-72-7 1 78-83-1 1	Propenol, 2,3-dibromo-, phosphete (3:1) Propenol, 2-methyl- (I,T)	
F027 J235 J140 J002	126-72-7 1 78-83-1 1 67-64-1 2	+Topenot, 2-methyl- (LT) -Propenone (I)	
F027 /235 /140	126-72-7 1 78-83-1 1 67-64-1 2 79-06-1 2	+Topenot, 2-methyl- (LT)	

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#### **Environmental Protection Agency**

Haz- ardous waste No.	Chemical abstracts No.	Substance	
U009	107-13-1	2-Propenentrile	
U152	126-98-7	2-Proponenitrile, 2-methyl- (I,T)	
U008	79-10-7		
U113	140-86-5	A man contact that the contact of th	
U118 U162	97-63-2 80-62-6	1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	
U194	107-10-8		
U083	78-87-5		
U148	123-33-1	3,6-Pyridezinedione, 1,2-dihydro-	
U196	110-86-1	Pyridine Pyridine 2 method	
U191 U237	109-06-8 66-75-1	Pyridine, 2-methyl- 2,4-(1H,3th)-Pyrimidinedione, 5-(bis(2- chloroethylemino)-	,
U164	58-04-2	4(1H)-Pyrimidinane, 2.3-dihydro-6-methyl-2-thicac-	
U189	930-55-2	Pyrrelidine, 1-nitroso-	
U209	50-65-5	Recerpine	
U201 U202	108-46-3 181-07-2	Resorcingl Saccharin, & salts	
U203	94-59-7	Safroie	
U204	7783-00-8	Solonique acid	
U204	7783-00-5	Solenium dioxide	
U205	7400-56-4	Selenium sulfide	
U205 U015	7486-56-4 115-02-6	Scienium sutifice SoS <sub>1</sub> (R,T) L-Serina, diaznecetate (ester)	
See F027	93-72-1	Silver (2,4,5-TP)	
U206	18883-88-4	Streptozotocin	
U103	77-78-1	Sulturic acid, dimethyl cotor	
U188	1314-80-3	Sulfur phosphide (Pt)	
See F827 U297	93-76-5 95-94-3	2.4.5-T	
U296	630-20-6	1.1.1.2-Tetrachierosthene	-
U200	79-34-5	1,1,2,2-Tetrachioroethane	
U210	127-18-4	Tetrachlereothylene	
See FB27	58-99-2	2,3,4,6-Tetrachlorephenel	
J213	109-98-9	Totrahydrofuran (i)	
U214	563-66-8	Thellium(i) acotate	
J215	6533-73-0	Theffjum(i).carbonate	
U216	7791-12-0	Thelium(i) chloride	
U216 U217	7791-12-8 10102-45-1	Theilium chloride Ticl     Theilium(i) nitrate	
U218	62-55-5	Thioscetamide	
J153	74-93-1	Thiomethanol (I,T)	
J244	137-28-8	Thisperoxydicarbenic diamide [(H,MC(S)] <sub>1</sub> S <sub>1</sub> , tetrumethyl-	
J219 J244	62-58-6 137-26-8	Thiourea. Thiram	
J220	108-68-3	Tokene	
J221	25376-45-8	Toluenediamine	
J223	26471-62-5	Toluene discovarete (R,T)	
J328 J353	95-53-4	o-Totaldino	
J222	106-48-0 636-21-5	p-Takudine a-Takudine hydrochloride	
JB11	61-82-5	1H-1,2,4-Trianol-3-emine	
J227	79-00-5	1,1,2-Trichlargethene	
J228	79-01-6	Trichioroethylene	•
J121 See F027	75-69-4 95-95-4	Trichleromonafluoremethiene 2,4,5-Trichlerophenel	
F927	88-06-2	2.4.6-Trichlarophenol	
J234	98-35-4	1,3,5-Trinitrobenzane (R,T)	
J182	123-63-7	1,3,5-Trioxano, 2,4,6-trimothyl-	
J235 J236	126-72-7	Tris(2,3-dibromopropyl) phosphate	
J236 J237	72-57-1 68-75-1	Trypen blue Unacil musterd	
J178		- Urea: N-ethyl-N-nitrese.	
J177	684-93-5		

Haz- ardous waste No.	Chemical abstracts No.	Substance
U248	1 81-81-2	Werfarin, & salts, when present at concentrations of 0.3% or less
U239	1330-20-7	Xviene (I)
U200	50-55-5	
U249	1314-84-7	Zinc phosphide Zn <sub>b</sub> P <sub>z</sub> , when present at concentrations of 10% or less

<sup>1</sup> CAS Number given for parent compound only.

(Approved by the Office of Management and Budget under control number 2050-0047) [45 FR 78529, 78541, Nov. 25, 1980, as amended at 46 FR 27477, May 20, 1981; 49 FR 19923, May 10, 1984; 50 FR 2000, Jan. 14, 1985; 50 FR 28744, July 15, 1985; 50 FR 42942, Oct. 23, 1985; 51 FR 6541, Feb. 25, 1986; 51 FR 10175, Mar. 24, 1986; 51 FR 28298, Aug. 6, 1986; 52 FR 21306, June 5, 1987; 52 FR 26012, July 10, 1987; 53 FR 13383, 13384, Apr. 22, 1988]

# APPENDIX I—REPRESENTATIVE SAMPLING METHODS

The methods and equipment used for sampling waste materials will vary with the form and consistency of the waste materials to be sampled. Samples collected using the sampling protocols listed below, for sampling waste with properties similar to the indicated materials, will be considered by the Agency to be representative of the waste.

Extremely viscous liquid—ASTM Standard D140-70 Crushed or powdered material—ASTM Standard D346-75 Soil or rock-like material—ASTM Standard D420-69 Soil-like material—ASTM Standard D1452-65

Fly Ash-like material—ASTM Standard D2234-76 (ASTM Standards are available from ASTM, 1916 Race St., Philadelphia, PA 19103)

Containerized liquid wastes—"COLIWASA" described in "Test Methods for the Evaluation of Solid Waste, Physical/Chemical Methods," b U.S. Environmental Protection Agency, Office of Solid Waste, Washington, D.C. 20460. [Copies may be obtained from Solid Waste Information, U.S. Environmental Protection Agency, 26 W. St. Clair St., Cincinnati, Ohio 45268]

Liquid waste in pits, ponds, lagoons, and similar reservoirs.—"Pond Sampler" described in "Test Methods for the Evaluation of Solid Waste, Physical/Chemical Methods." \*\*

This manual also contains additional information on application of these protocols.

#### A. Extraction Procedure (EP)

1. A representative sample of the waste to be tested (minimum size 100 grams) shall be obtained using the methods specified in Appendix I or any other method capable of yielding a representative sample within the meaning of Part 260. IFor detailed guidance on conducting the various aspects of the EP see "Test Methods for the Evaluation of Solid Waste, Physical/Chemical Methods" (incorporated by reference, see § 260.11).]

2. The sample shall be separated into its component liquid and solid phases using the method described in "Separation Procedure" below. If the solid residue a obtained using this method totals less than 0.5% of the original weight of the waste, the residue can be discarded and the operator shall treat the liquid phase as the extract and proceed immediately to Step 8.

(weight of pad + solid) - (tare weight of pad) ×100 initial weight of sample

3. The solid material obtained from the Separation Procedure shall be evaluated for its particle size. If the solid material has a surface area per gram of material equal to, or greater than, 3.1 cm² or passes through a 9.5 mm (0.375 inch) standard sieve, the operator shall proceed to Step 4. If the surface area is smaller or the particle size larger than specified above, the solid material shall be prepared for extraction by crush-

APPENDIX II—EP TOXICITY TEST PROCEDURES

These methods are also described in "Samplers and Sampling Procedures for Hazardous Waste Streams," EPA 600/2-80-018, January 1980.

The percent solids is determined by drying the filter pad at 80°C until it reaches constant weight and then calculating the percent solids using the following equation:

Percent solids =